



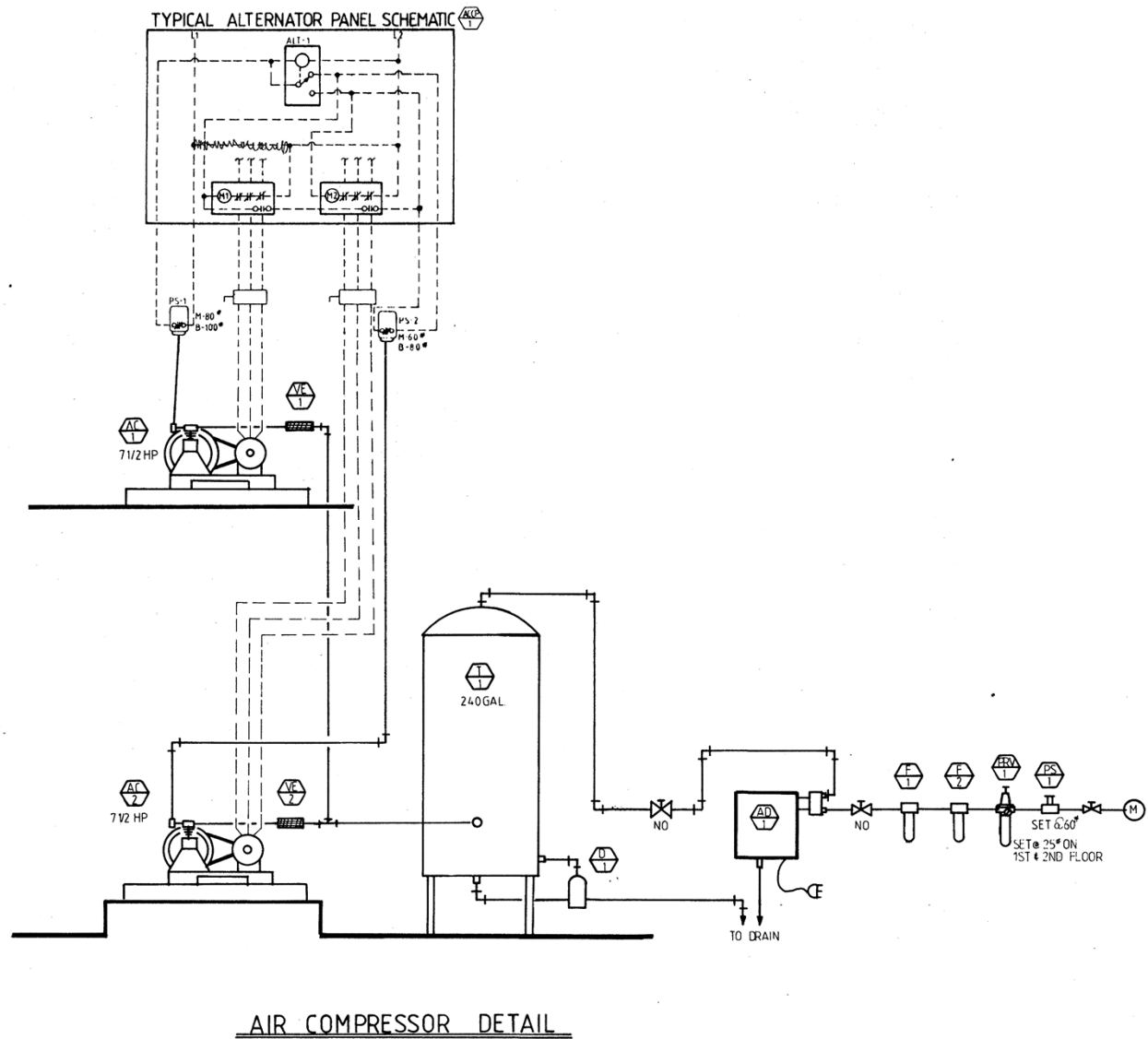
# A Field Perspective on Pneumatic Control and Actuation Systems

## Control Air Systems



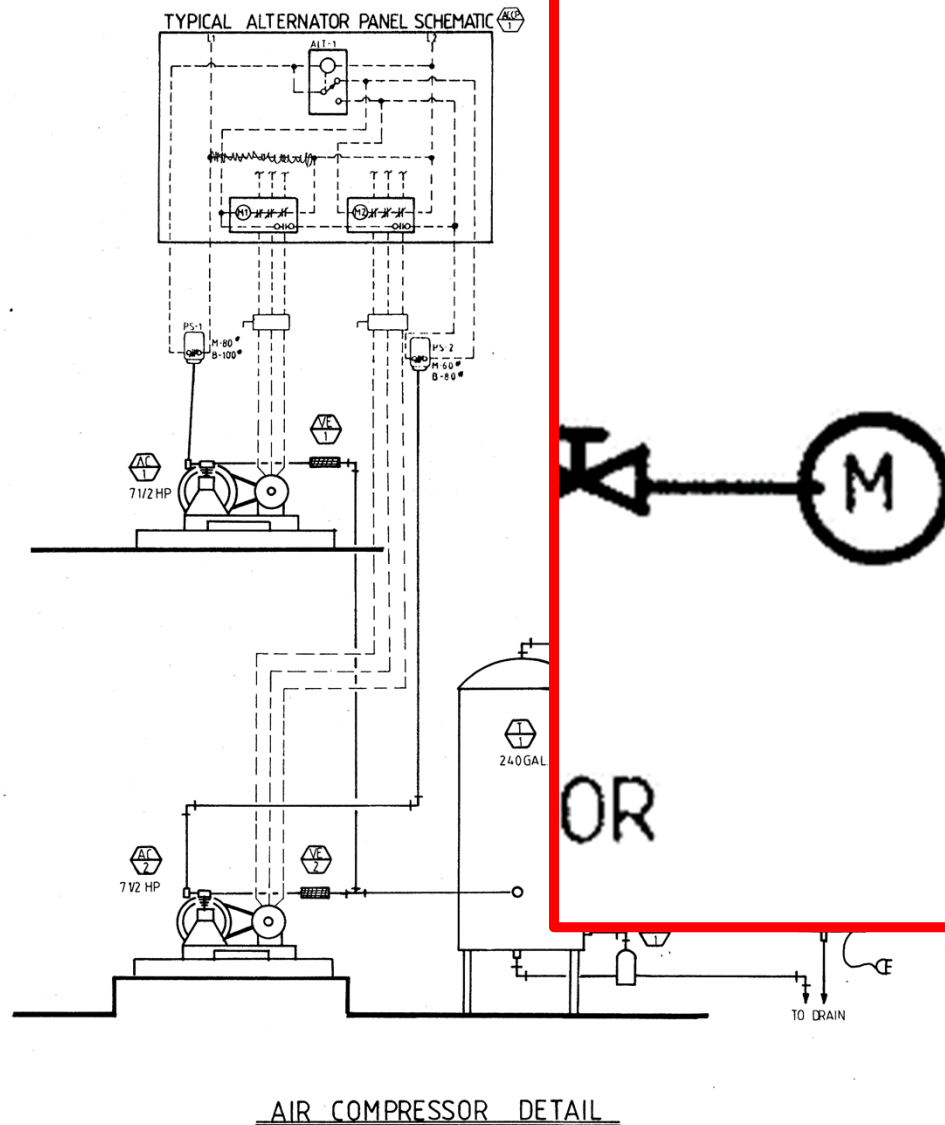
Presented By:  
David Sellers  
Senior Engineer, Facility Dynamics Engineering

# A Typical Pneumatic Air Supply System



# A Typical Pneumatic Air Supply System

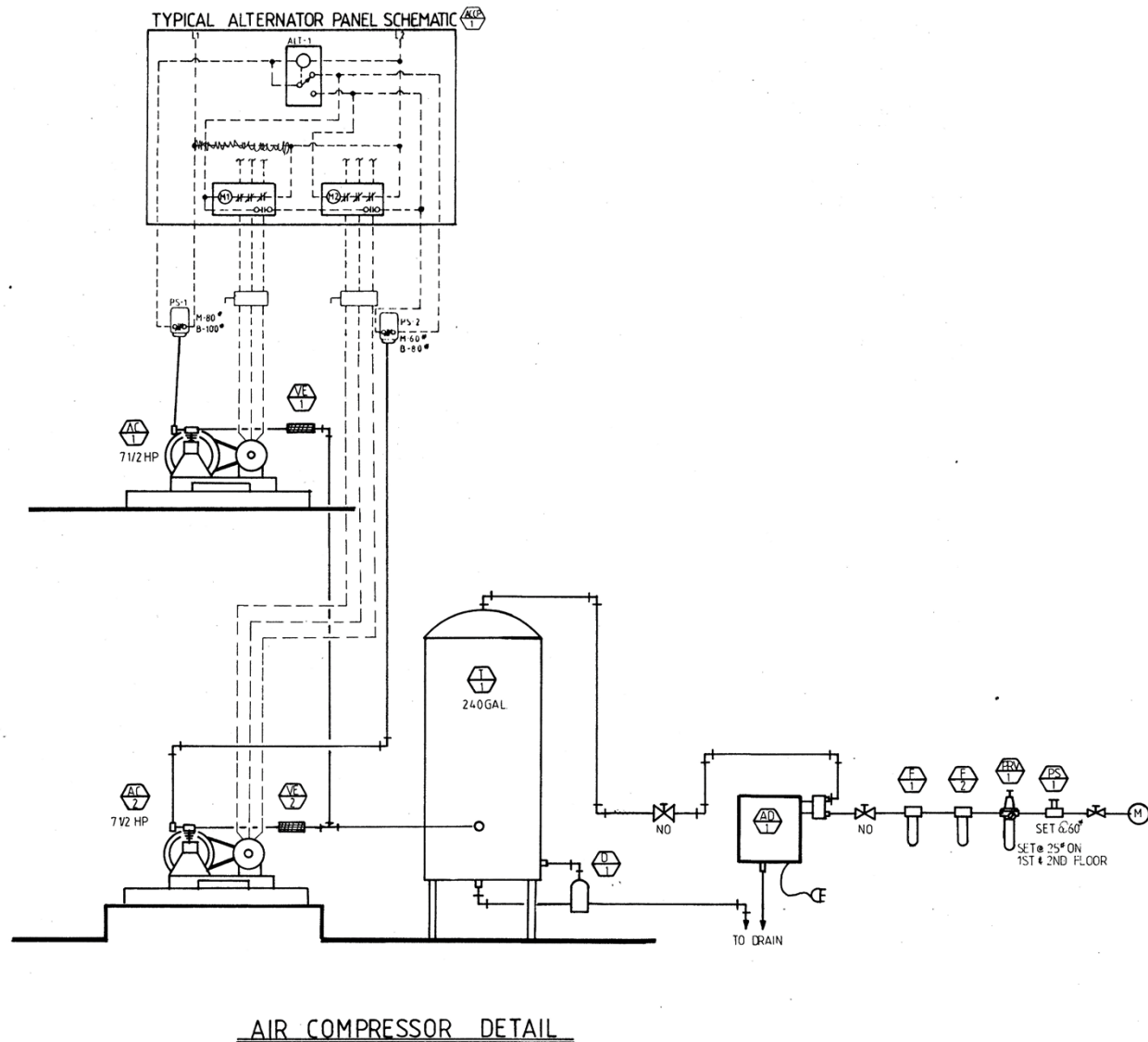
“Main Air”



# A Typical Pneumatic Air Supply System

## Key Points

1. Pneumatics are truly “plug and play” Jay Santos



# Pneumatics are “Plug and Play” Jay Santos

You could use this  
transmitter ...



# Pneumatics are “Plug and Play” Jay Santos

You could use this transmitter ...



... with this controller ...



# Pneumatics are “Plug and Play” Jay Santos

You could use this transmitter ...



... with this controller ...



... and this set point adjuster ...



# Pneumatics are “Plug and Play” Jay Santos

You could use this transmitter ...



... with this controller ...



... and this set point adjuster ...



... and this interlock ...





# Pneumatics are “Plug and Play” Jay Santos

You could use this transmitter ...



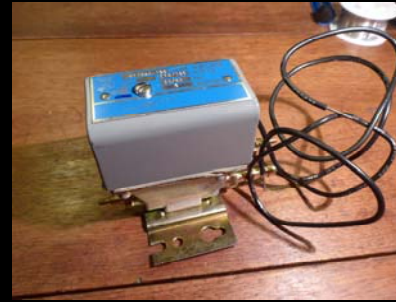
... with this controller ...



... and this set point adjuster ...



... and this interlock ...



... and this final control element ....



# Pneumatics are “Plug and Play” Jay Santos

You could use this transmitter ...



... with this controller ...



... and this set point adjuster ...



... and this interlock ...



... and this final control element ....



... and it would work just fine

- Physical layer
  - Copper or poly tubing is supported
  - Barbed, compression, soldered, or brazed fittings are supported
- Communications layer
  - 3-15 psig range is standard
  - Anything any range from 0 – 25 psig is supported for signal transmission and sequencing
  - 25 psig power is standard for first tier controllers, 50 psig and higher is supported for distribution and positioning applications and process control

# Taking a Look at the “Physical Layer”

- Copper or poly tubing is supported

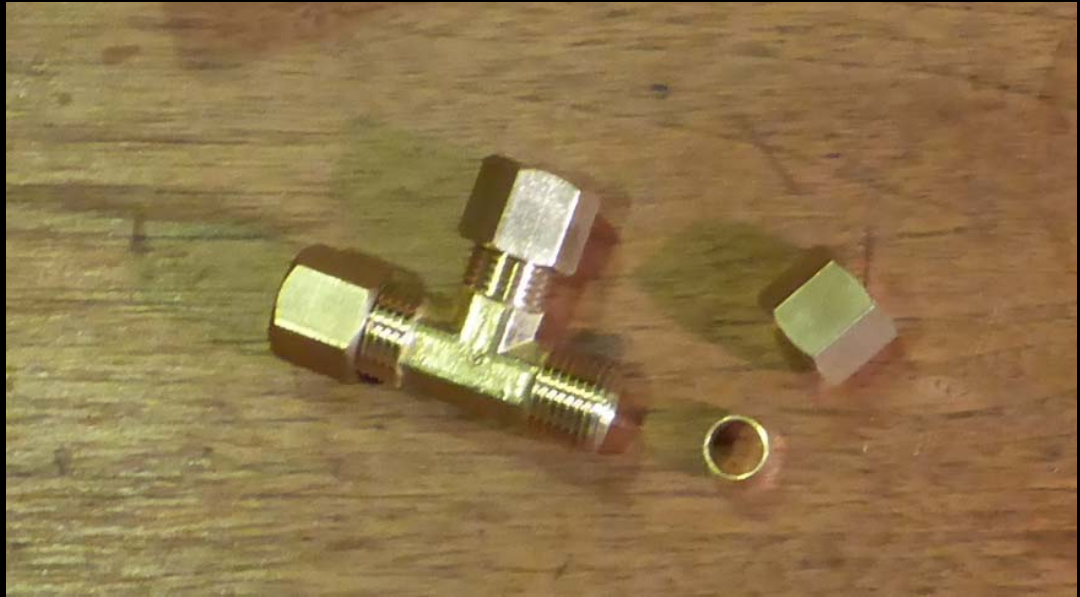


# Taking a Look at the “Physical Layer”

- Copper or poly tubing is supported
- Barbed, compression, soldered, or brazed fittings are supported







### Material and Tubing Specifications

Specifications for PLEXCO™ 2600 Instube® Pneumatic Control Tubing

Property	Test Method	Typical Value
<b>Flame Retardant Compound</b>		
Melt Index	ASTM D 1238	0.6 ± 0.1 g/10 min
Density	ASTM D 792	1.1 ± 0.005 g/cm <sup>3</sup>
Tensile Strength—ultimate (20 in./min.)	ASTM D 638	> 2000 psi (> 1.8 MPa)
Tensile Elongation—ultimate (20 in./min.)	ASTM D 638	> 600 %
Bending Modulus	ASTM D 747	30,000 ± 5,000 psi (20639 ± 34.5 MPa)
Shore A Hardness	ASTM D 1700	97 ± 3
Shore D Hardness	ASTM D 1700	45 ± 3
Water Absorption	ASTM D 570	5% maximum
Polyethylene Classification	ASTM D 3350	PE 11 or PE 12
Stress-Crack Resistance	ASTM D 1693	> 200 hours non-failure
Brittleness Temperature	ASTM D 745	< -104.8°F (< -76°C)
<b>Pneumatic Instrument Control Tubing</b>		
Burst Pressure	ASTM D 1599	5/32" — > 500 psi (> 3.4 MPa) 1/4" — > 500 psi (> 3.4 MPa) 3/8" — > 500 psi (> 3.4 MPa) 1/2" — > 350 psi (> 3.4 MPa)
Minimum Bend Radius	—	5/32" — 0.50 in. (13 mm) 1/4" — 0.75 in. (19 mm) 3/8" — 1.50 in. (38 mm) 1/2" — 1.88 in. (48 mm)
Maximum Allowable Pulling Load During Installation	—	5/32" — 15 lb. (66 N) 1/4" — 33 lb. (147 N) 3/8" — 76 lb. (338 N) 1/2" — 106 lb. (472 N) 5/32" Twintube — 30 lb. (133 N) 1/4" Twintube — 65 lb. (289 N)
Flammability	UL 94 UL 910	V-2 UL Classified (NFPA 90A)
Flame Propagation	UL 1820	< 5 ft (< 1.5 m)
Smoke Density — Peak Optical Density	UL 1820	< 0.5
Smoke Density — Average Optical Density	UL 1820	< 0.15

**NOTICE** — This table provides typical physical property information for polyethylene compounds used to manufacture DRISCOPEX™ 2600 tubing products. It is intended for comparing compounds and tubing. It is not a product specification, and it does not establish minimum or maximum values or manufacturing tolerances for compounds or tubing. The typical property values for compound were determined using compression-molded plaques prepared from compound. Values obtained from tests of specimens taken from tubing can vary from these typical values. Performance Pipe has made every reasonable effort to ensure the accuracy of this information, but this table may not provide all necessary information, particularly with respect to special or unusual applications. This information may be changed from time to time without notice. Contact Performance Pipe to determine if you have the most recent edition.

# Taking a Look at Poly Ratings

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### Material and Tubing Specifications

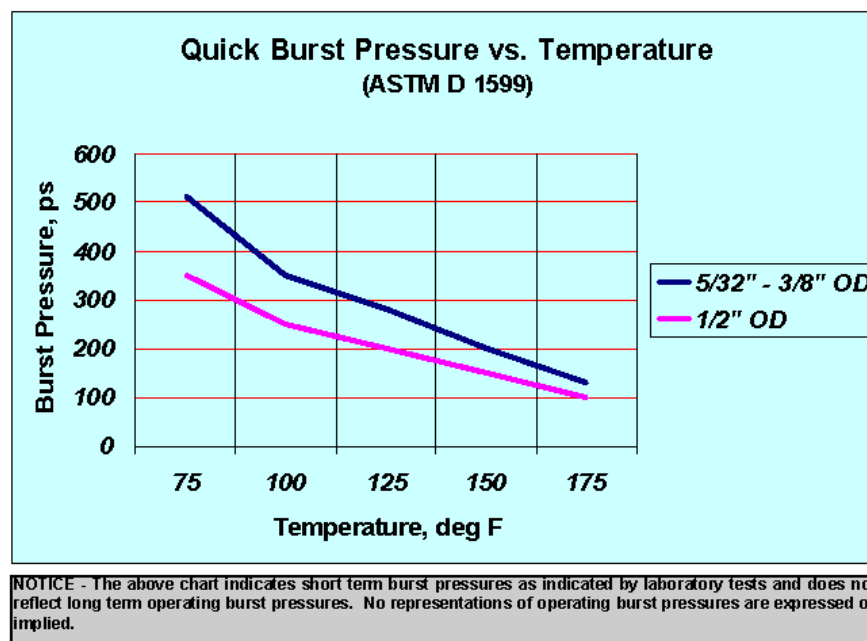
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# Taking a Look at Poly Ratings





## Plastic Pipes and Maximum Operating Temperatures

ABS, PE, PVC, CPVC, PB, PP and SR - pressure and operating temperatures

Learn more

Replay

### Operating Temperature

- is the maximum operating temperatures for plastic pipe materials.

Plastic Pipe Material	Operating Temperature			
	With Pressure		Without Pressure	
	(°F)	(°C)	(°F)	(°C)
ABS - Acrylonitrilebutadiene Styrene	100	38	180	82
PE - Polyethylene	100	38	180	82
PVC - Polyvinylchloride	100	38	140	60
CPVC - Chlorinated Polyvinyl Chloride	180	82	180	82
PB - Polybutylene	180	82	200	93
PP - Polypropylene	100	38	180	82
SR - Styrene Rubber Plastic	150	66		

### Maximum Short Time Operating Temperature

- for pipes without pressure.

- PVC : 95 °C
- PP : 100 °C
- PE : 95 °C

### Heat Distortion Temperature

- is the temperature where a test piece of a material placed in a heat medium with a bending load ( $18.6 \text{ kg/cm}^2$ ) applied - reaches a specified deflection.

- ABS: 104 - 106 °C
- PVC: 54 - 80 °C
- HDPE: 43 - 49 °C
- LDPE: 32 - 41 °C
- PP: 57 - 64 °C

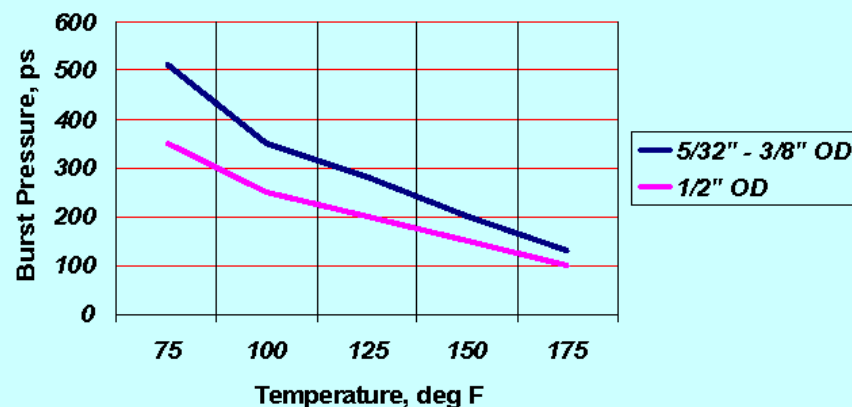
### Vicat Softening Temperature

- is the temperature where a needle shaped penetrator sinks into a test piece a specified depth when a specified vertical load (1 kg) is applied.

- ABS: 102.3 °C
- PVC: 92 °C

# Taking a Look at Poly Ratings

Quick Burst Pressure vs. Temperature  
(ASTM D 1599)



NOTICE - The above chart indicates short term burst pressures as indicated by laboratory tests and does not reflect long term operating burst pressures. No representations of operating burst pressures are expressed or implied.

## Operating Temperature

- is the maximum operating temperatures for plastic pipe materials.

Plastic Pipe Material	Operating Temperature			
	With Pressure		Without Pressure	
	(°F)	(°C)	(°F)	(°C)
ABS - Acrylonitrilebutadiene Styrene	100	38	180	82
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PVC - Polyvinylchloride	100	38	140	60
CPVC - Chlorinated Polyvinyl Chloride	180	82	180	82
PB - Polybutylene	180	82	200	93
PP - Polypropylene	100	38	180	82
SR - Styrene Rubber Plastic	150	66		

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- for pipes without pressure.

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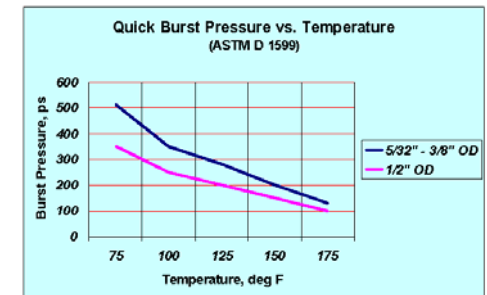
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# Taking a Look at Poly Ratings



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# The Same, But Different

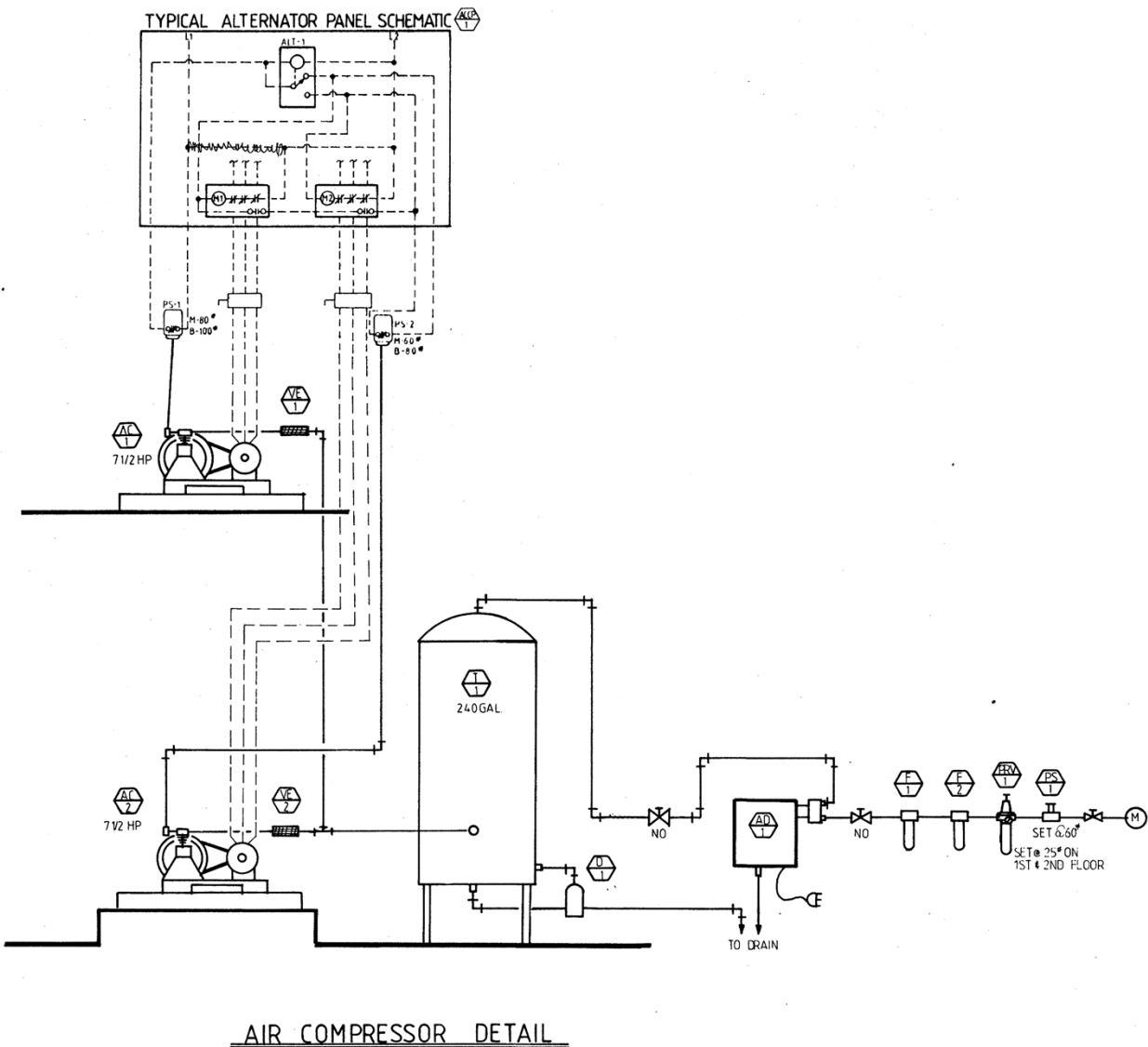
- What do these have in common?
- Functionally, what is different about them?



# A Typical Pneumatic Air Supply System

## Key Points

1. Pneumatics are truly  
“plug and play” Jay Santos





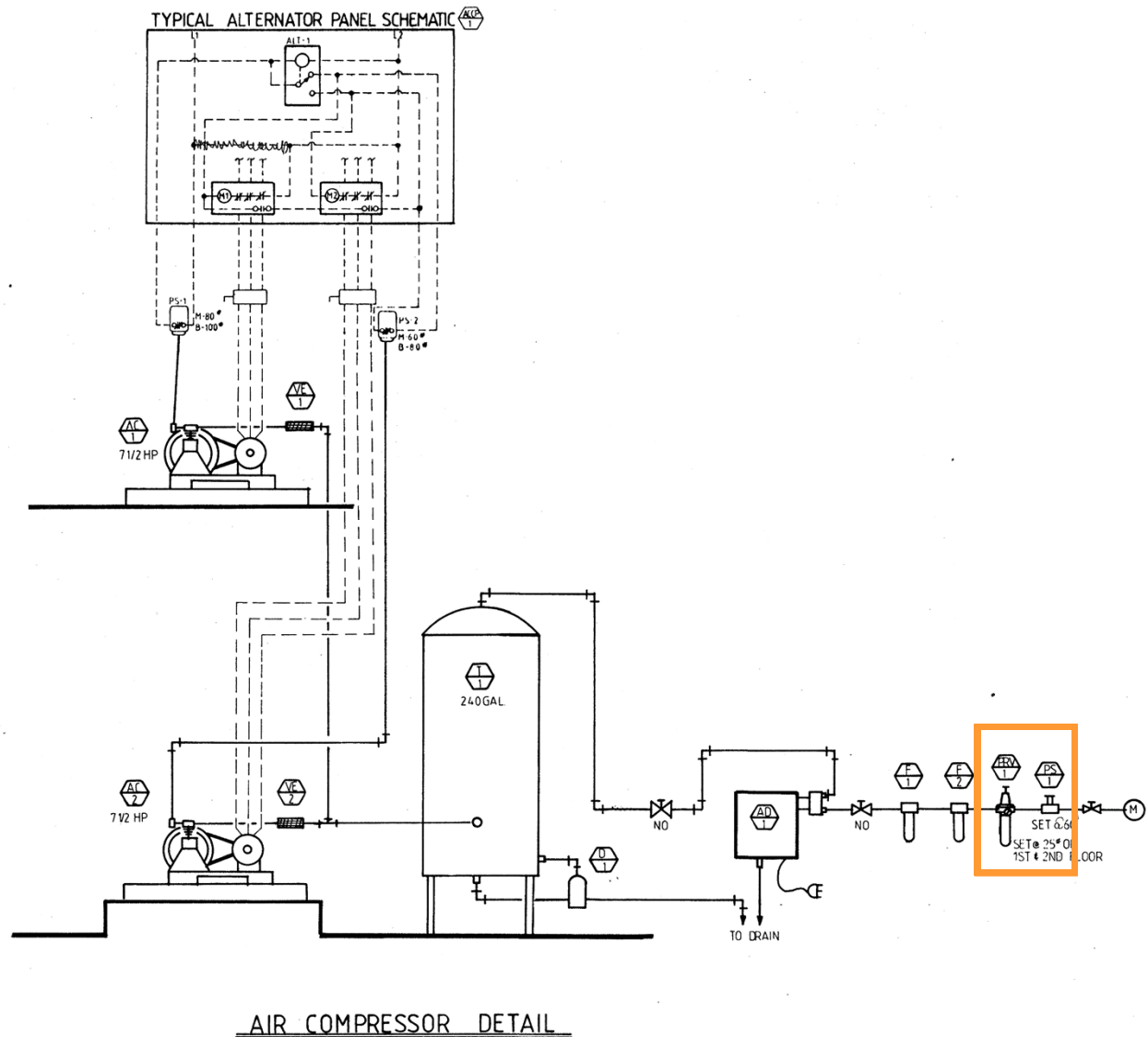




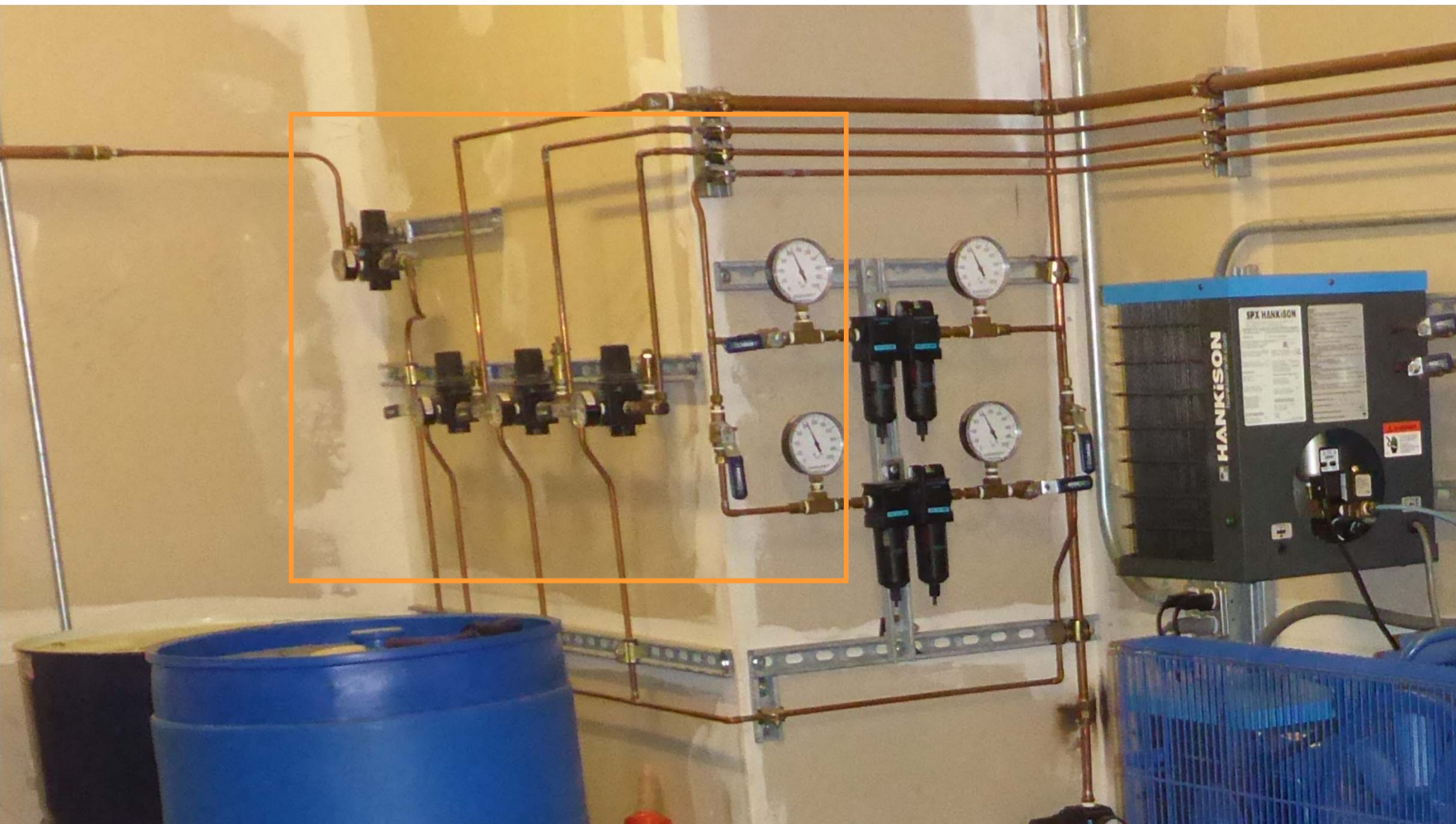
# A Typical Pneumatic Air Supply System

## Key Points

1. Pneumatics are truly “plug and play” Jay Santos
2. Pressure regulation is important





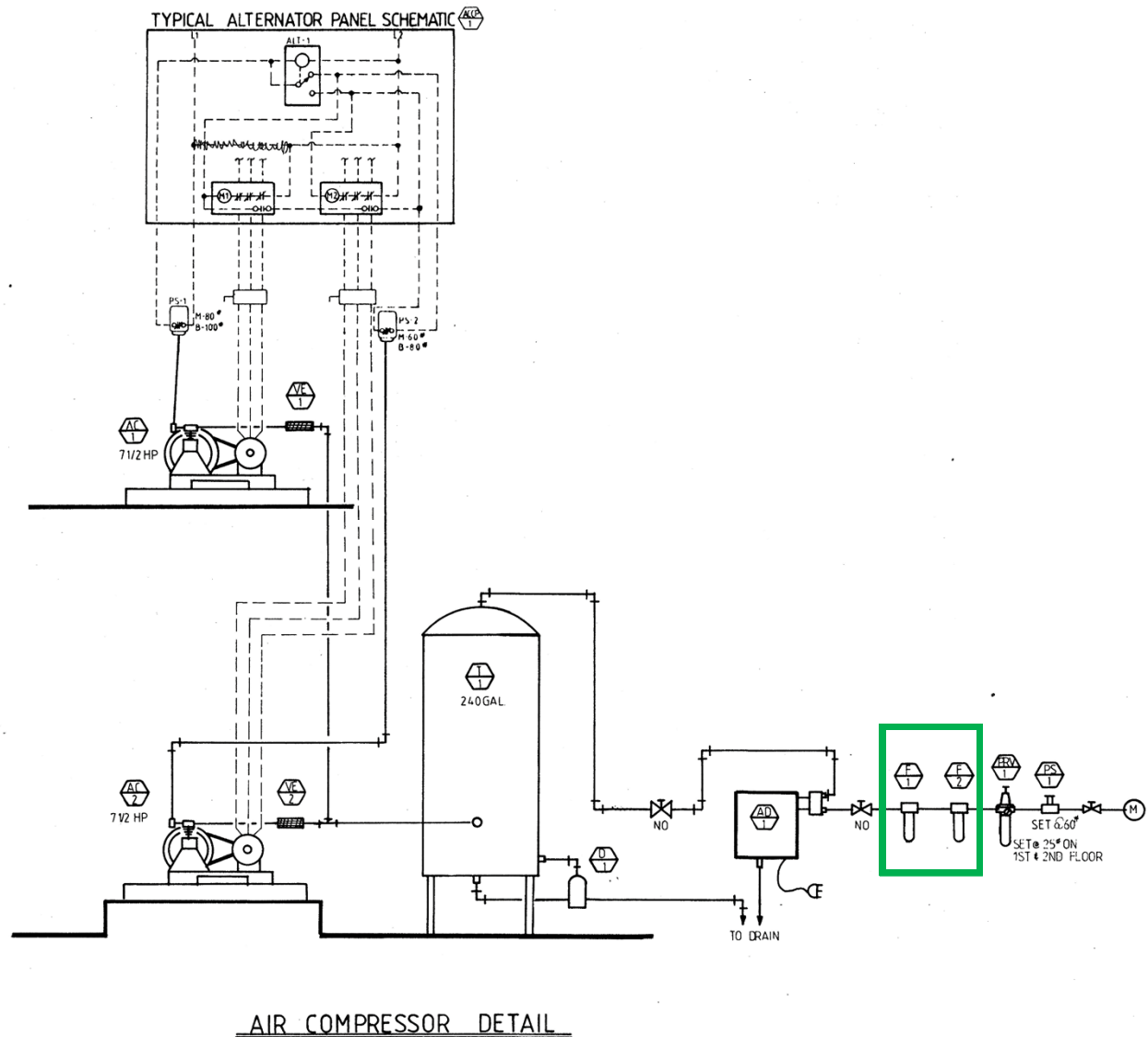


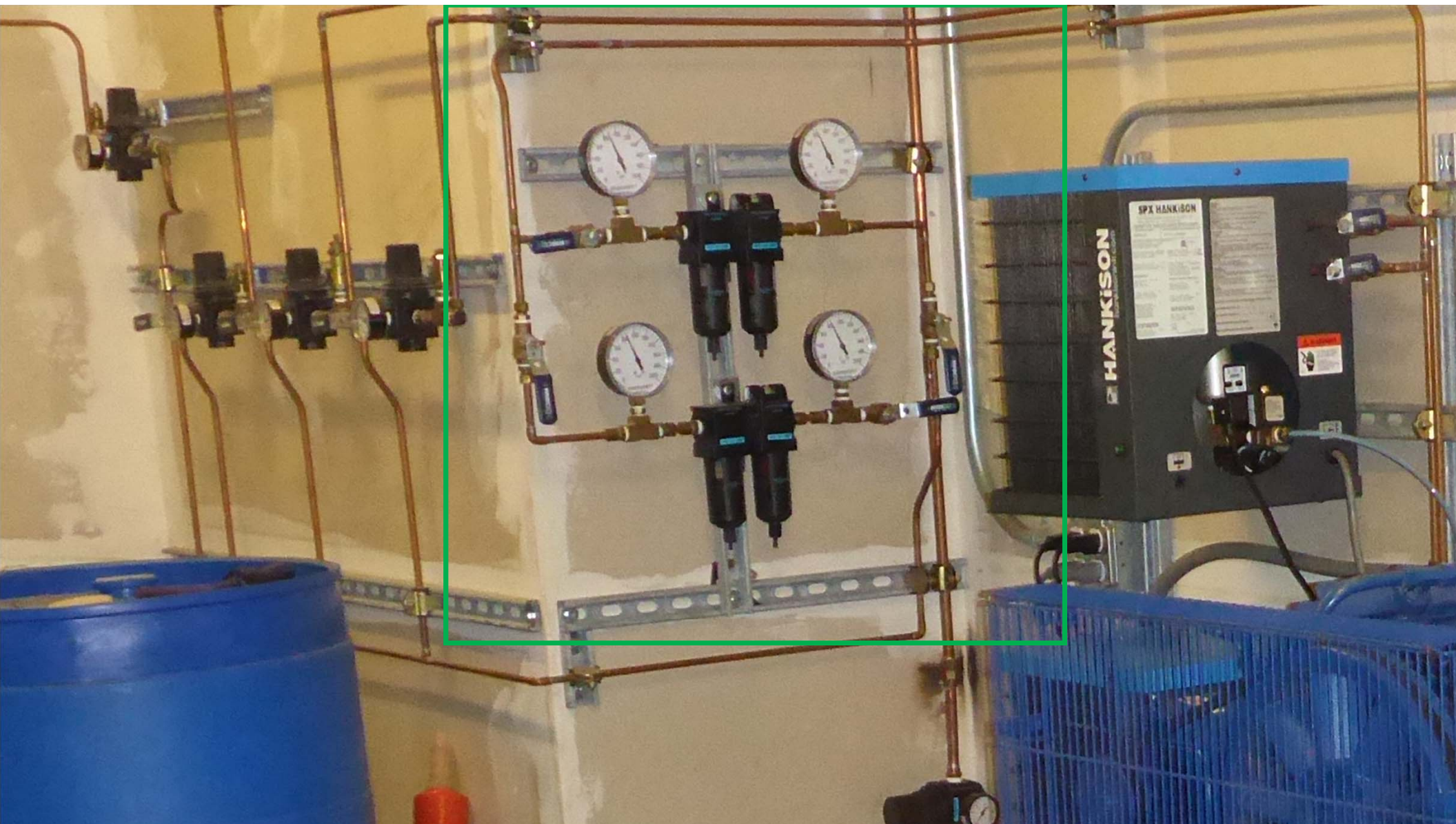


# A Typical Pneumatic Air Supply System

## Key Points

1. Pneumatics are truly “plug and play” Jay Santos
2. Pressure regulation is important
3. Clean, oil-free air is important

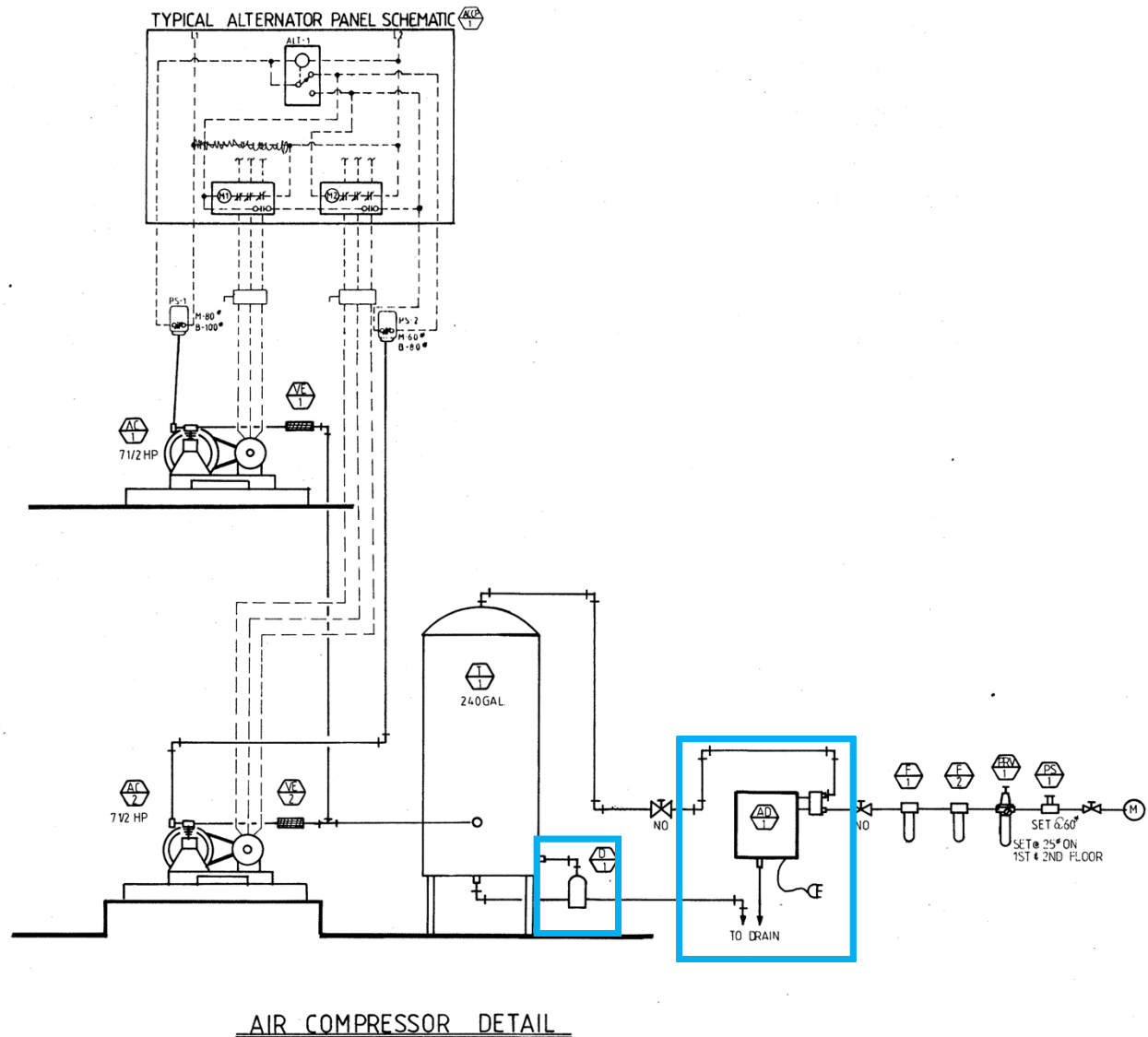




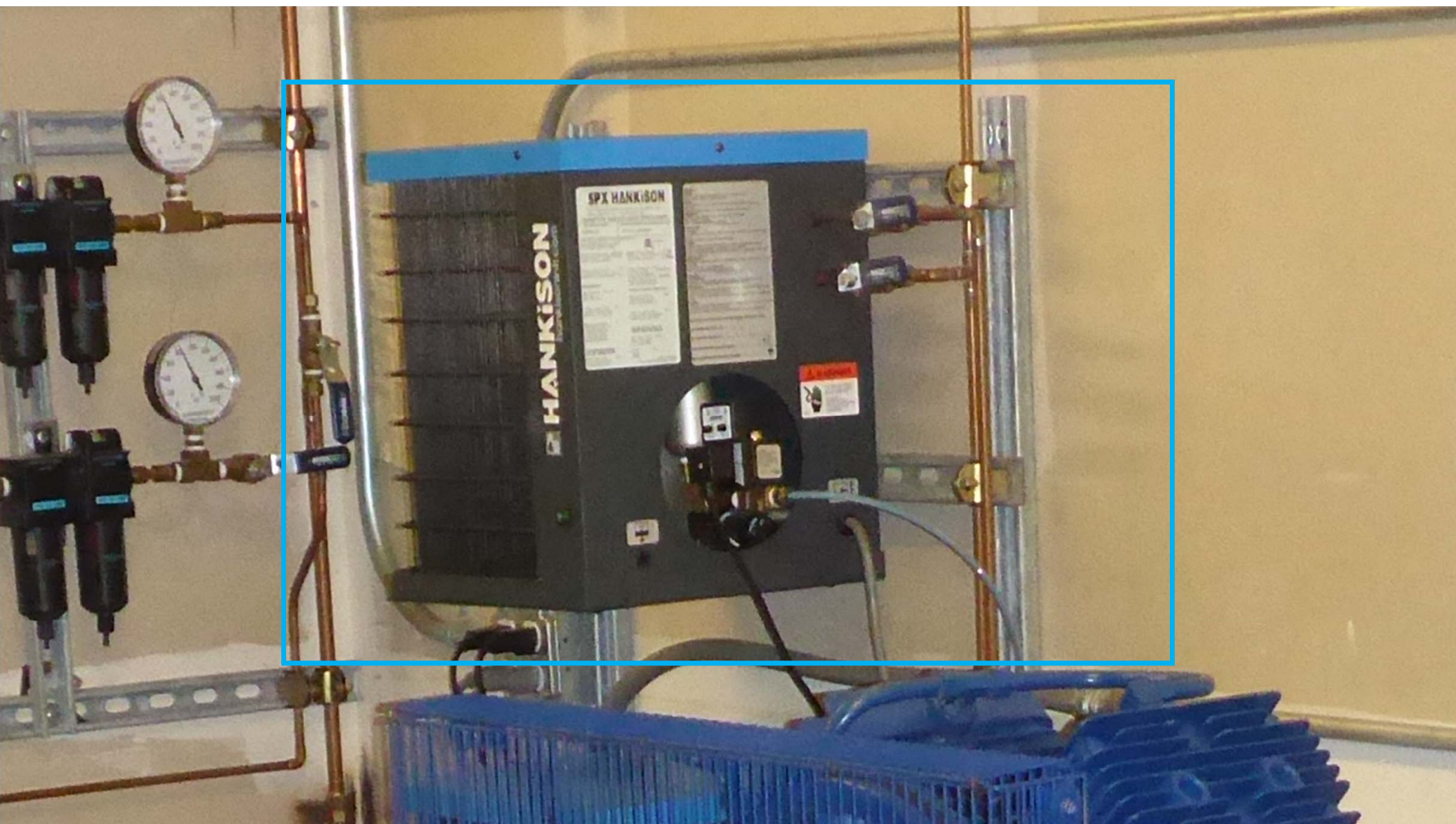
# A Typical Pneumatic Air Supply System

## Key Points

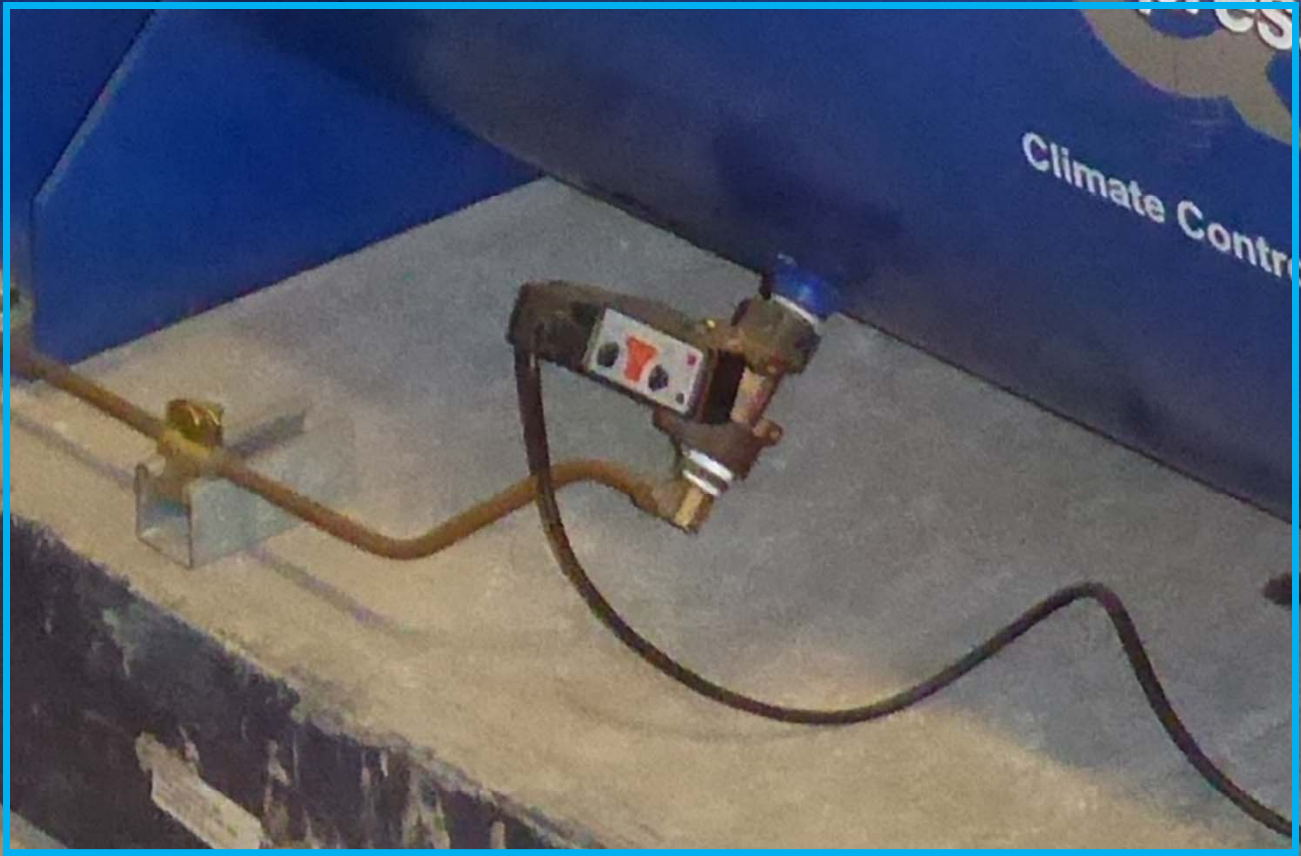
1. Pneumatics are truly “plug and play” Jay Santos
2. Pressure regulation is important
3. Clean, oil-free air is important
4. Dry air is important











Why is there  
Water in  
Compressed  
Air?





# Sea Level

ALTITUDE: SEA LEVEL  
BAROMETRIC PRESSURE: 29.921 in. HG  
ATMOSPHERIC PRESSURE: 14.696 psia

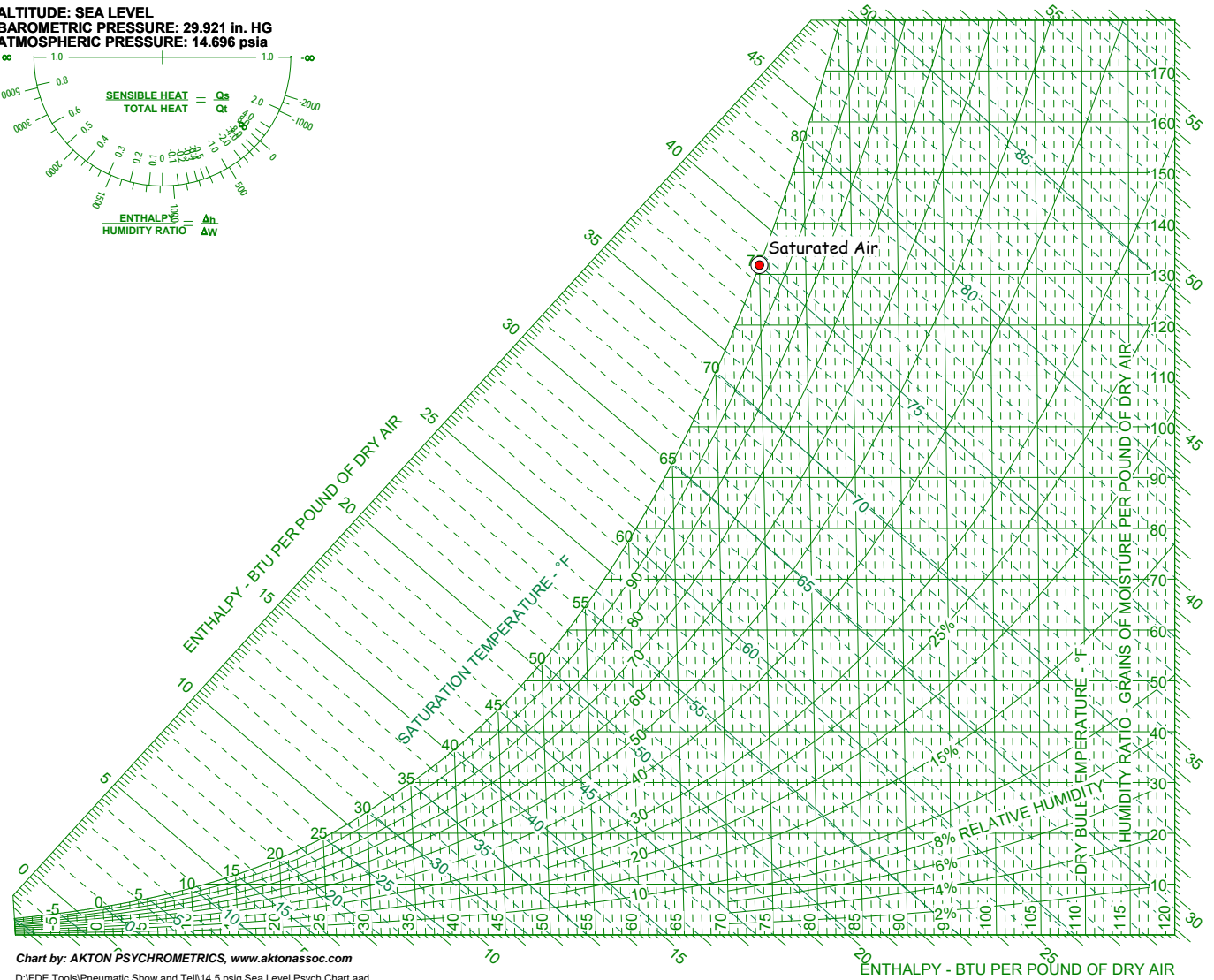
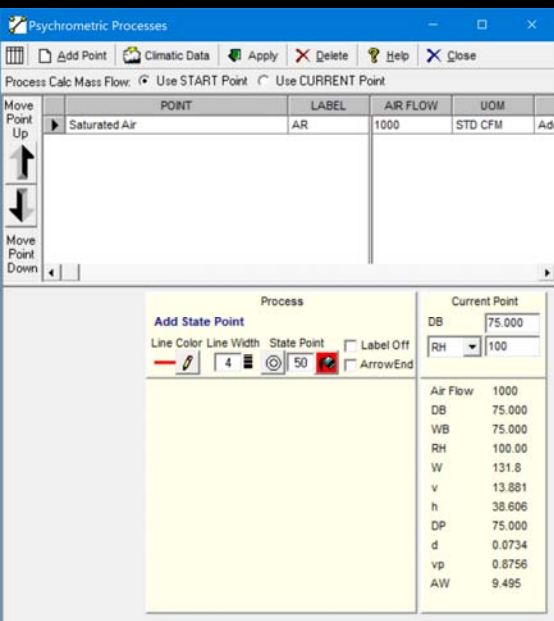
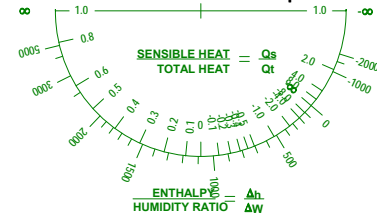
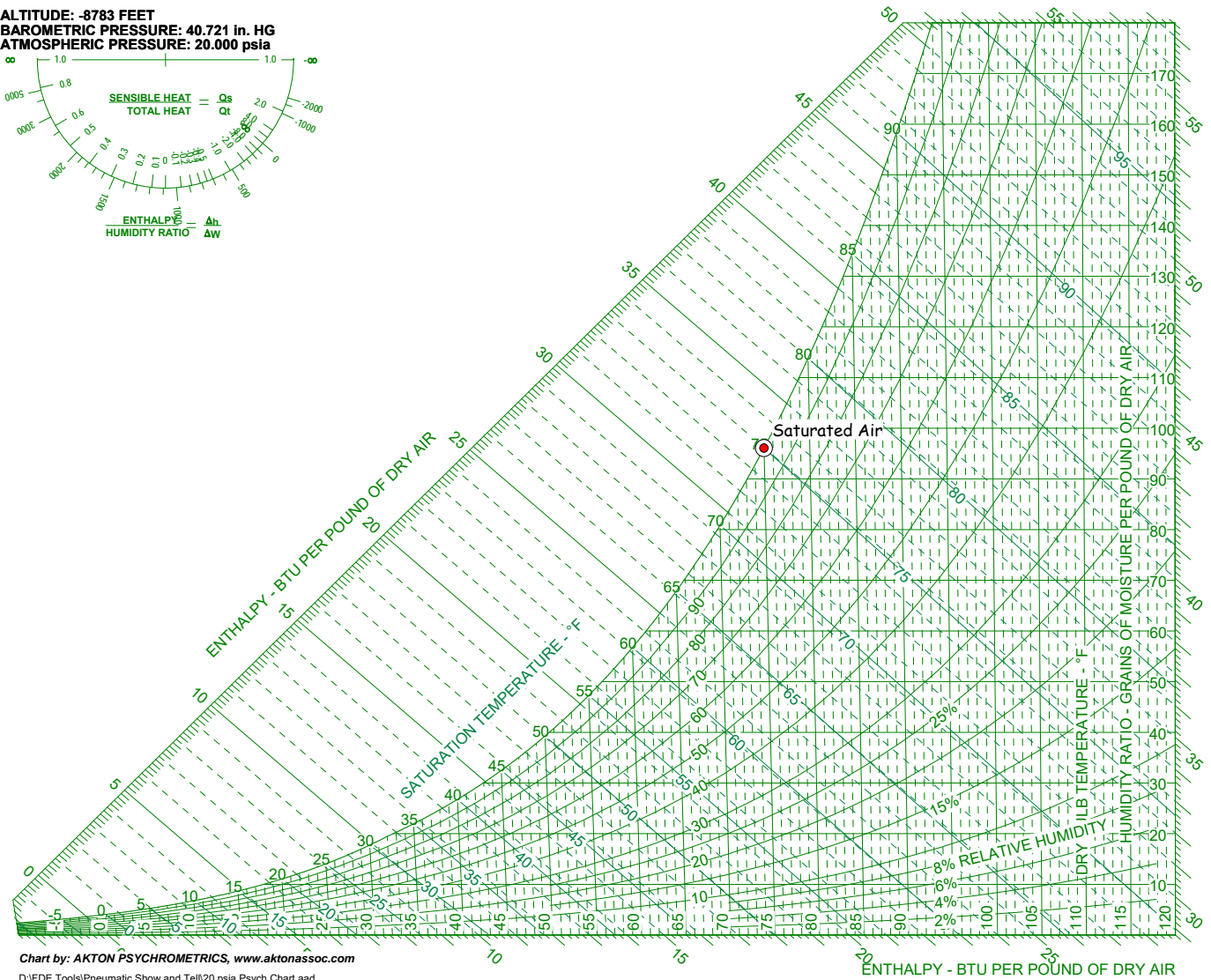
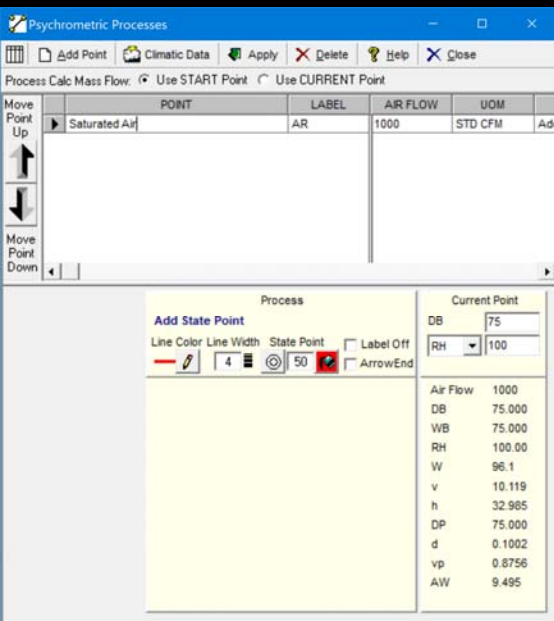
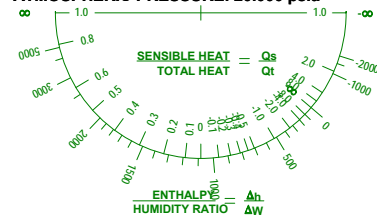


Chart by: AKTON PSYCHROMETRICS, [www.aktonassoc.com](http://www.aktonassoc.com)

D:\FDE Tools\Pneumatic Show and Tell\14.5 psig Sea Level Psych Chart.aad

# 20 psia

ALTITUDE: -8783 FEET  
BAROMETRIC PRESSURE: 40.721 in. HG  
ATMOSPHERIC PRESSURE: 20.000 psia





# 25 psia

ALTITUDE: -15472 FEET  
BAROMETRIC PRESSURE: 50.901 in. HG  
ATMOSPHERIC PRESSURE: 25.000 psia

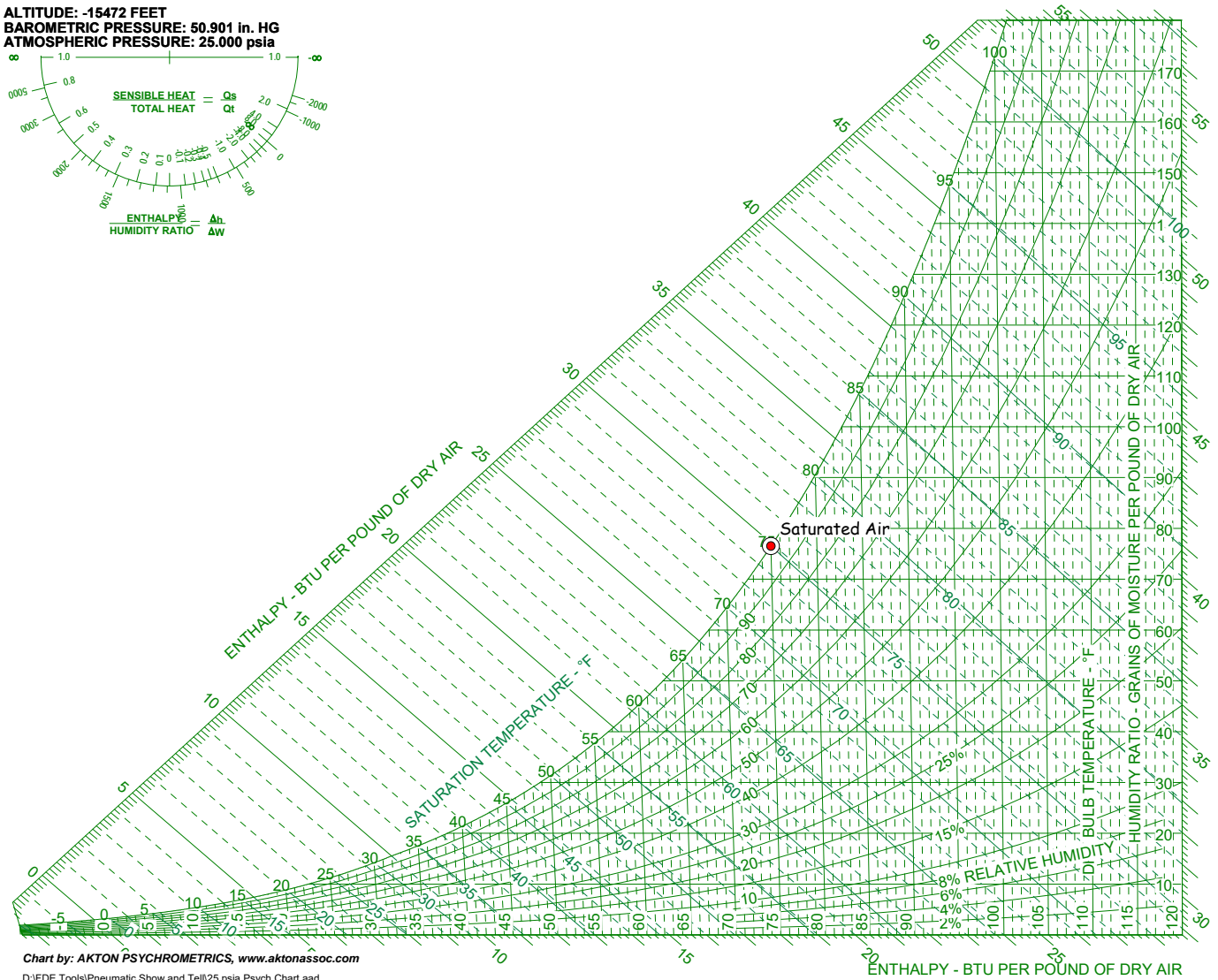
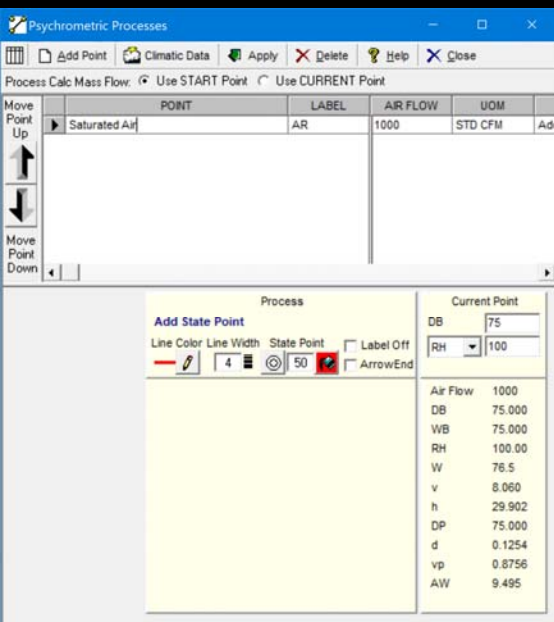
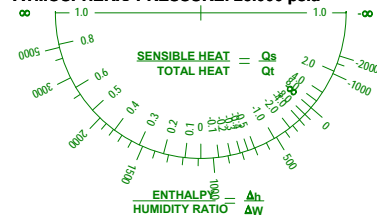
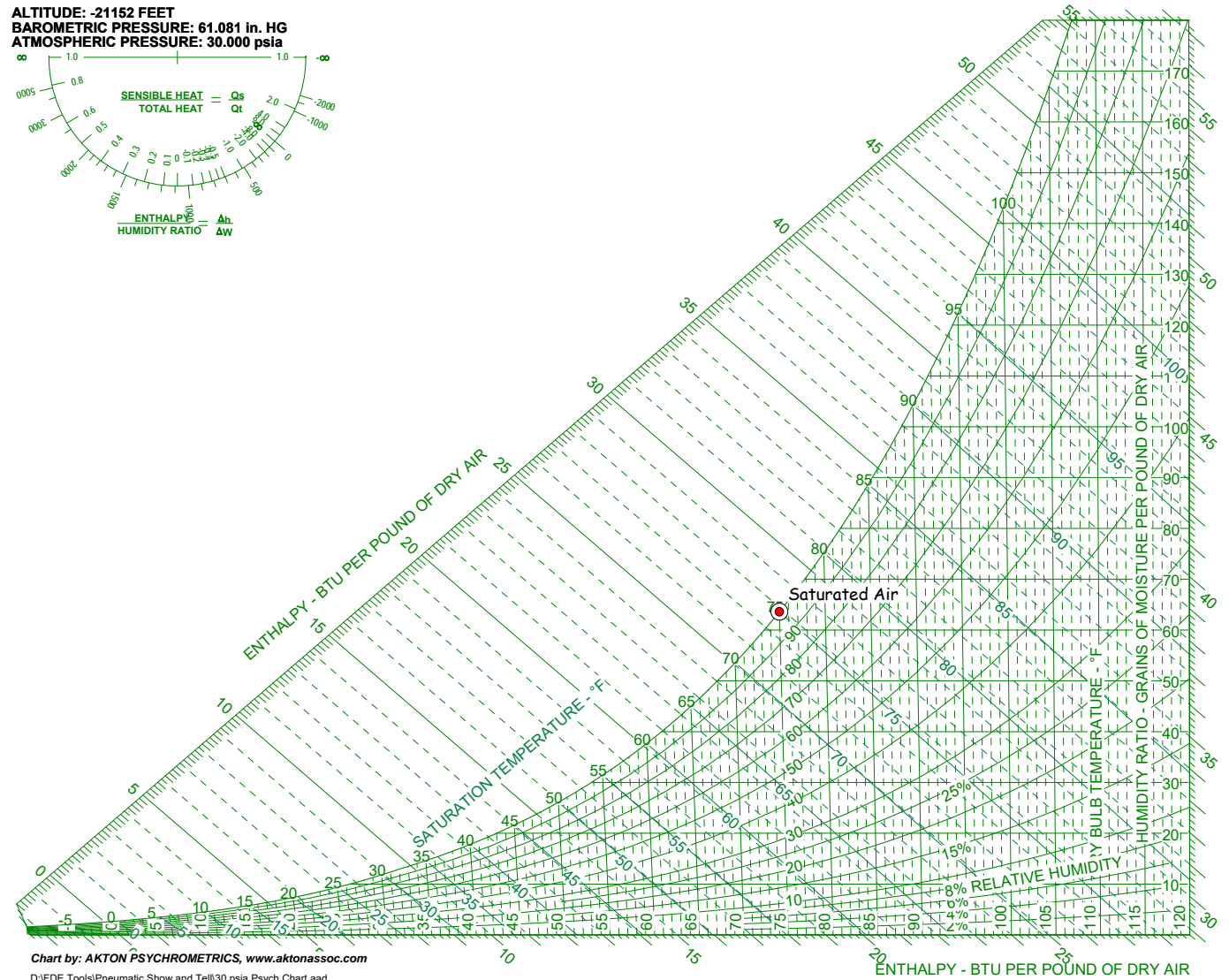
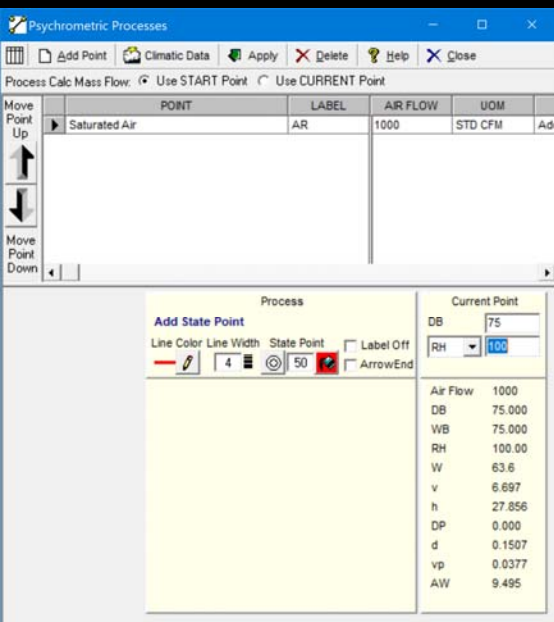
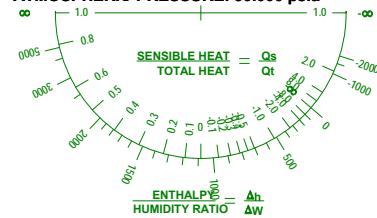


Chart by: AKTON PSYCHROMETRICS, [www.aktonassoc.com](http://www.aktonassoc.com)

D:\FDE Tools\Pneumatic Show and Tell\25 psia Psych Chart.aad

# 30 psia

ALTITUDE: -21152 FEET  
BAROMETRIC PRESSURE: 61.081 in. HG  
ATMOSPHERIC PRESSURE: 30.000 psia



# 35 psia

ALTITUDE: -26111 FEET  
BAROMETRIC PRESSURE: 71.262 in. HG  
ATMOSPHERIC PRESSURE: 35.001 psia

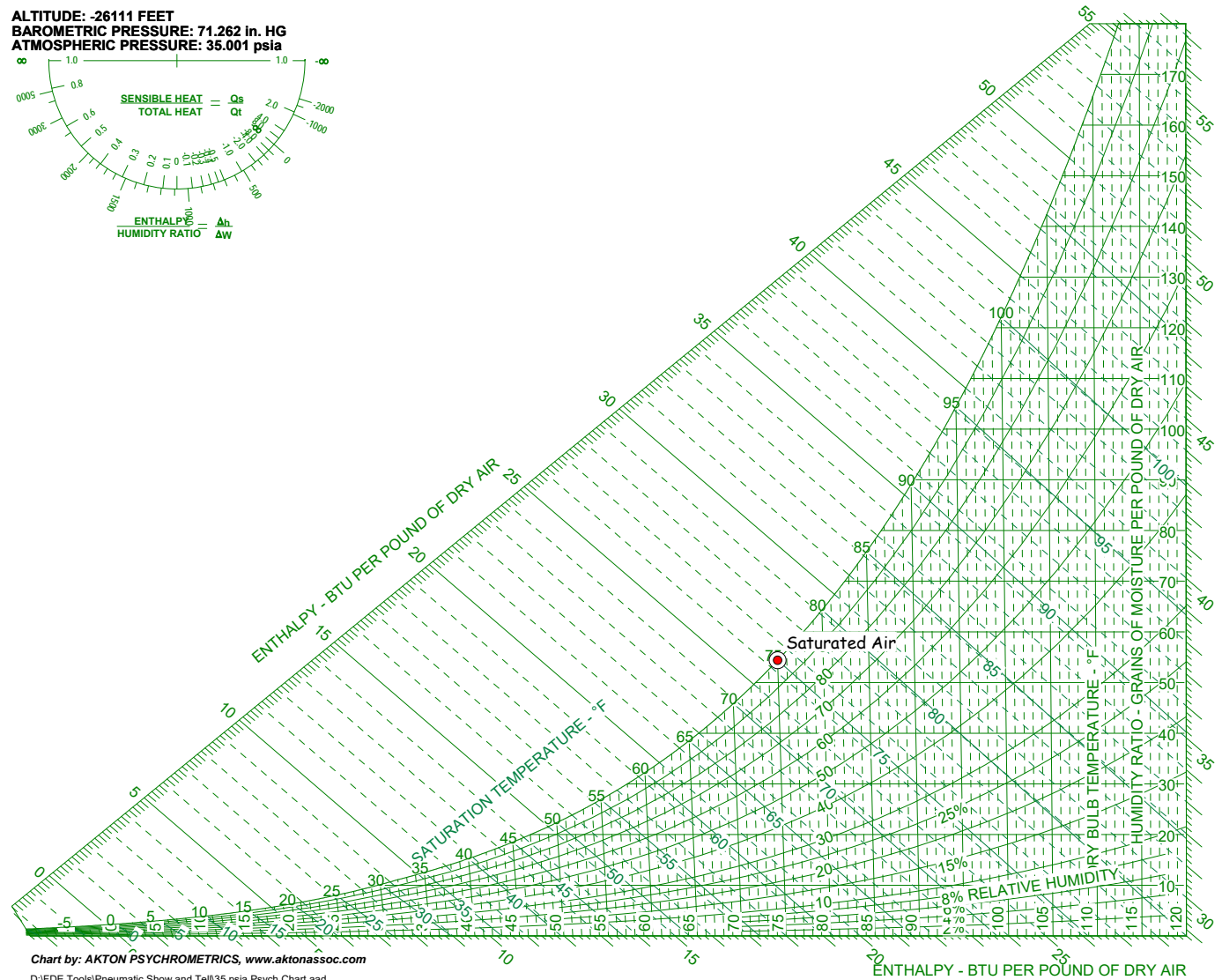
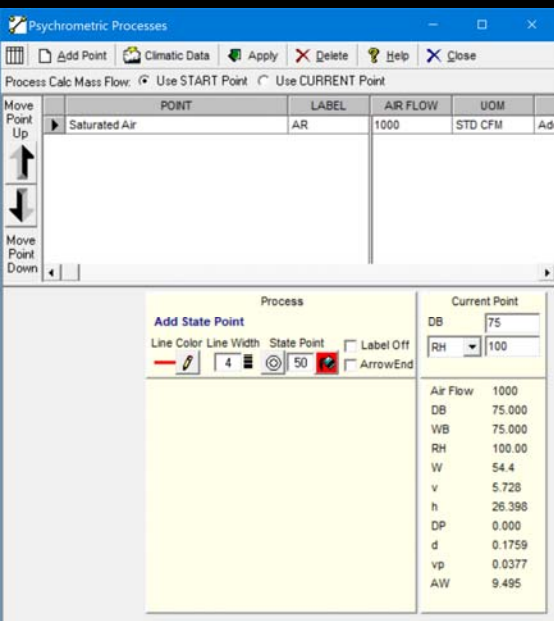
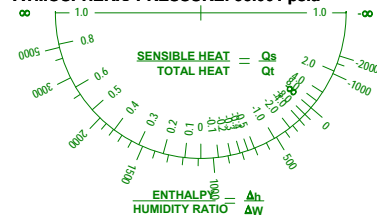
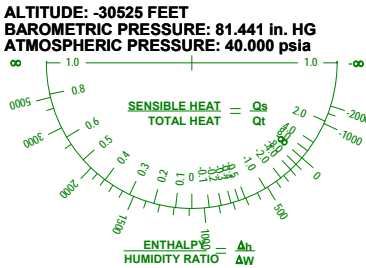


Chart by: AKTON PSYCHROMETRICS, [www.aktonassoc.com](http://www.aktonassoc.com)  
D:\FDE Tools\Pneumatic Show and Tell\35 psia Psych Chart.aad

40 psia



Psychrometric Processes

Process Calc Mass Flow: ☒ Use START Point ☐ Use CURRENT Point

POINT	LABEL	AIR FLOW	UOM
Saturated Air	AR	1000	STD CFM

Process

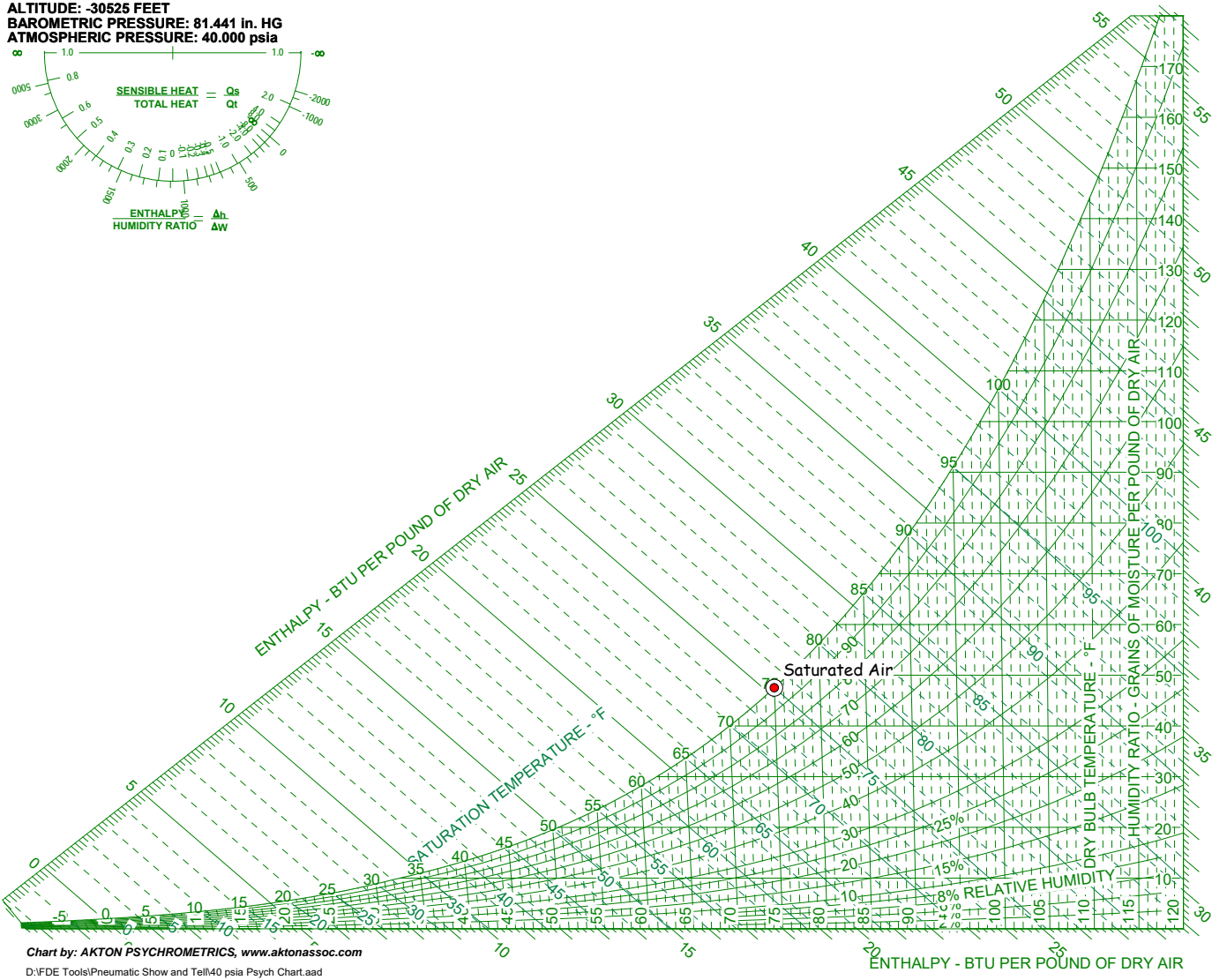
Add State Point

Line Color ☐ Line Width ☐ State Point ☐ Label Off ☐ ArrowEnd ☐

Current Point

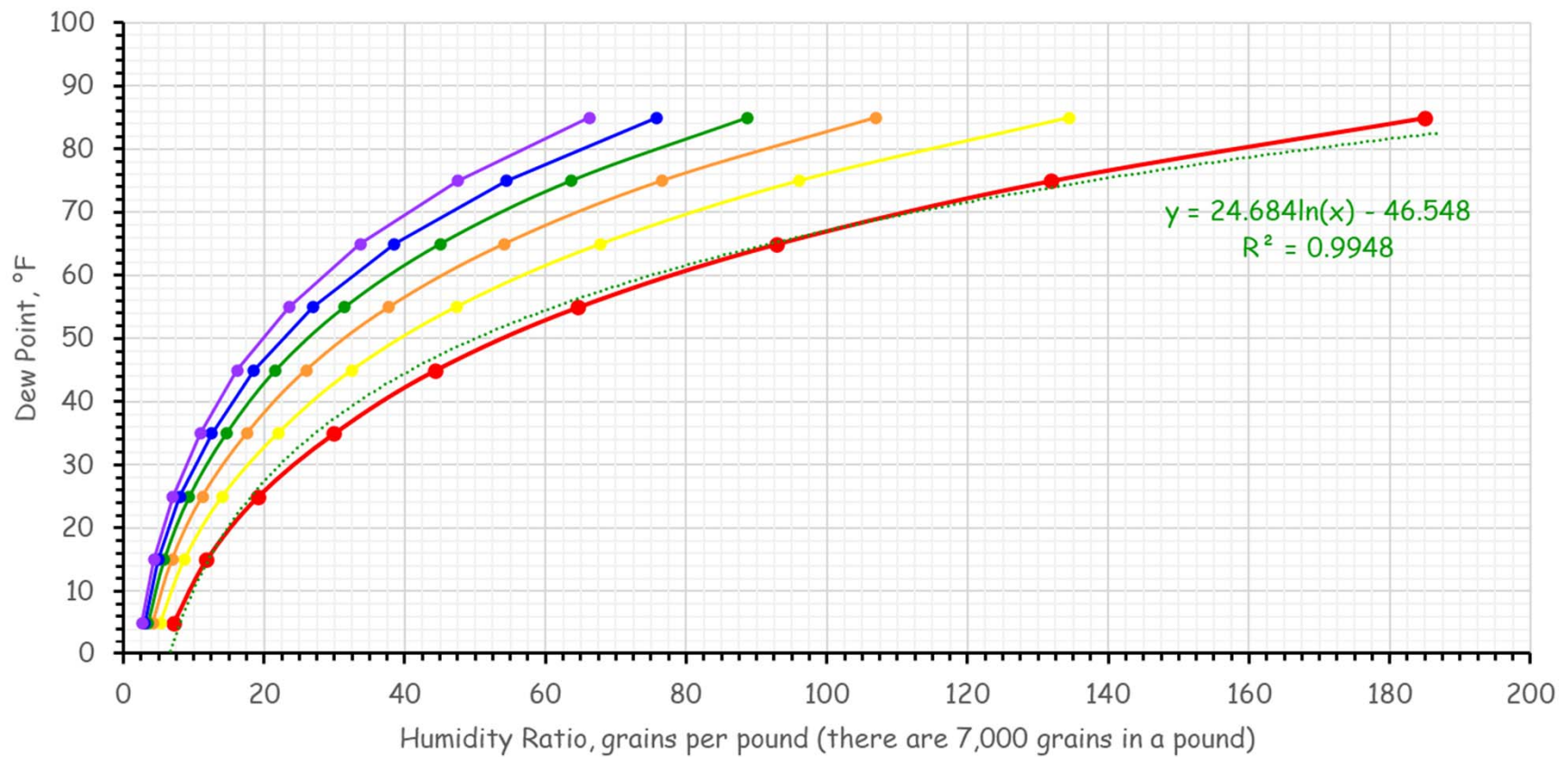
DB 75  
RH 100

Air Flow 1000  
DB 75.000  
WB 75.000  
RH 100.00  
W 47.5  
v 5.004  
h 25.306  
DP 0.000  
d 0.2012  
vp 0.0377  
AW 9.495





# Dew Point vs. Humidity Ratio at Different Pressures



- Dew Point vs. Humidity Ratio, 14.7 psia
- "Dew Point vs. Humidity Ratio - 25 psia"
- "Dew Point vs. Humidity Ratio - 35 psia"
- ..... Log. (Dew Point vs. Humidity Ratio, 14.7 psia)
- Dew Point vs. Humidity Ratio - 20 psia
- "Dew Point vs. Humidity Ratio - 30 psia"
- "Dew Point vs. Humidity Ratio - 40 psia"

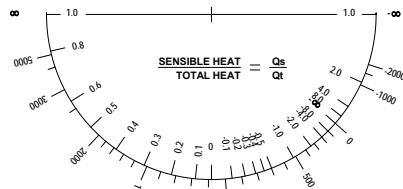


# PSYCHROMETRIC CHART NORMAL TEMPERATURE

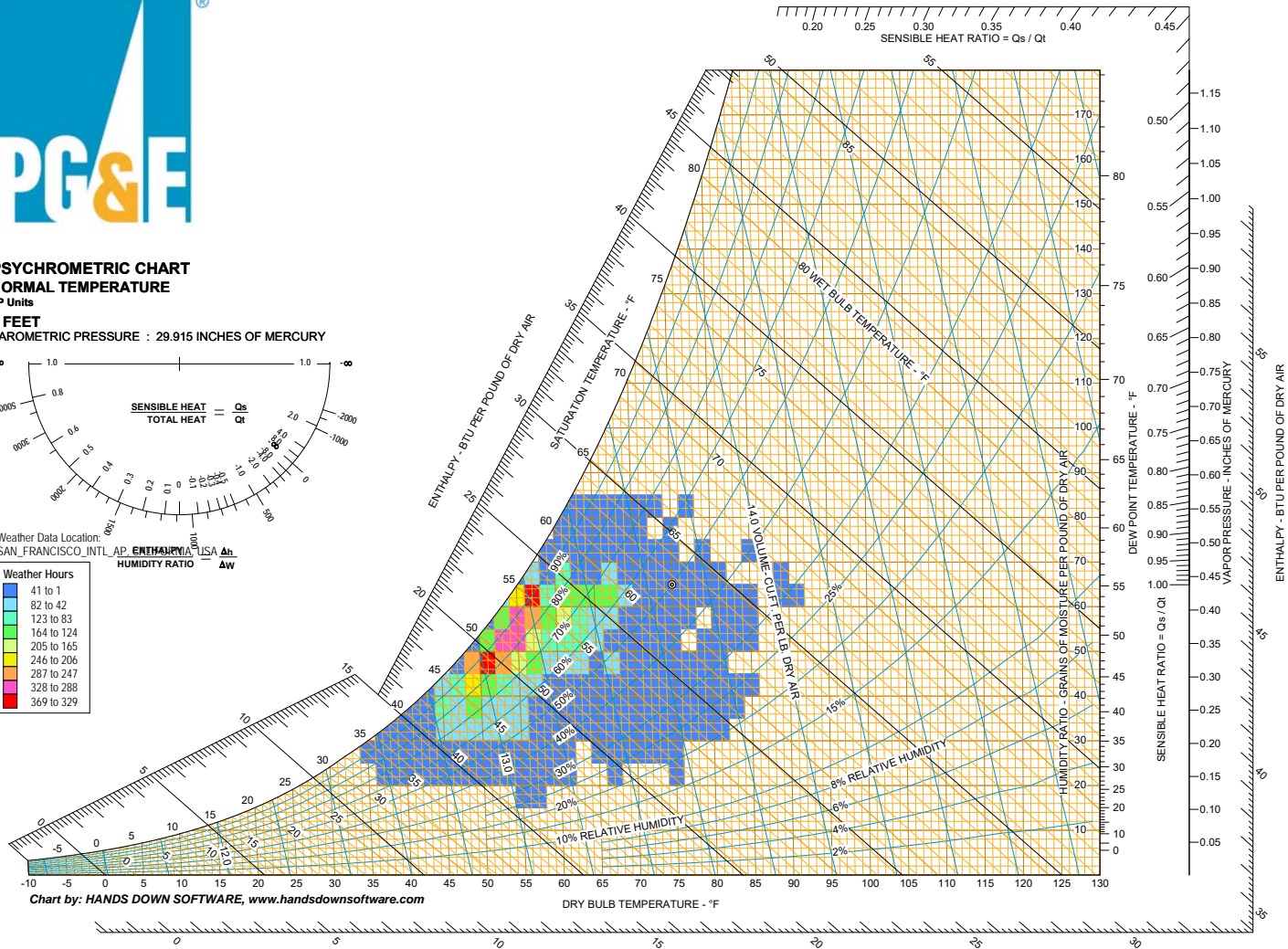
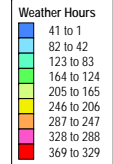
I-P Units

7 FEET

BAROMETRIC PRESSURE : 29.915 INCHES OF MERCURY

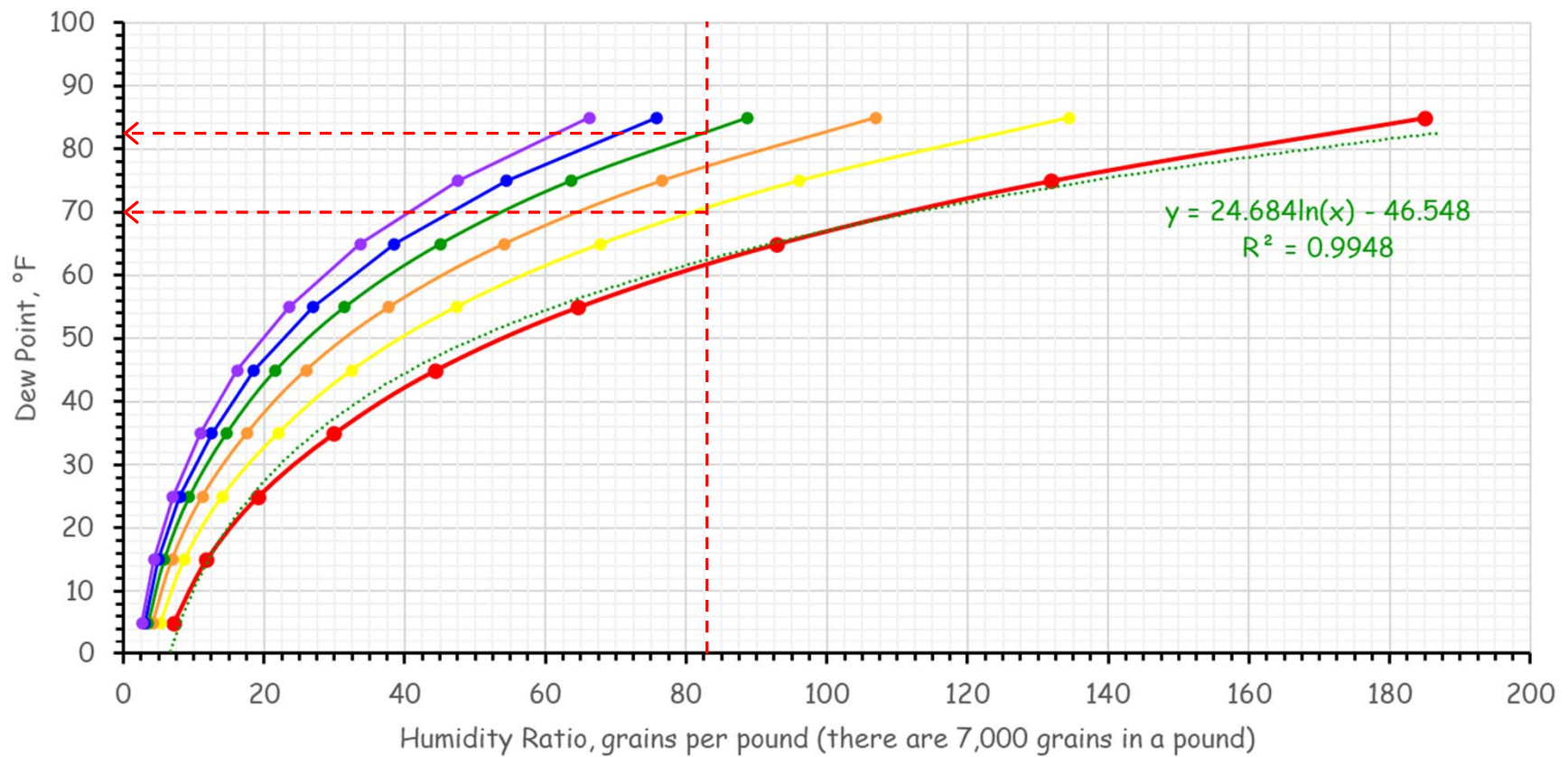


Weather Data Location:  
SAN FRANCISCO INTL AP, ENTHALPIA, USA



File Not Saved

# Dew Point vs. Humidity Ratio at Different Pressures



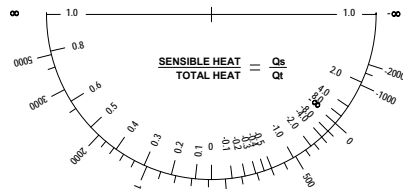


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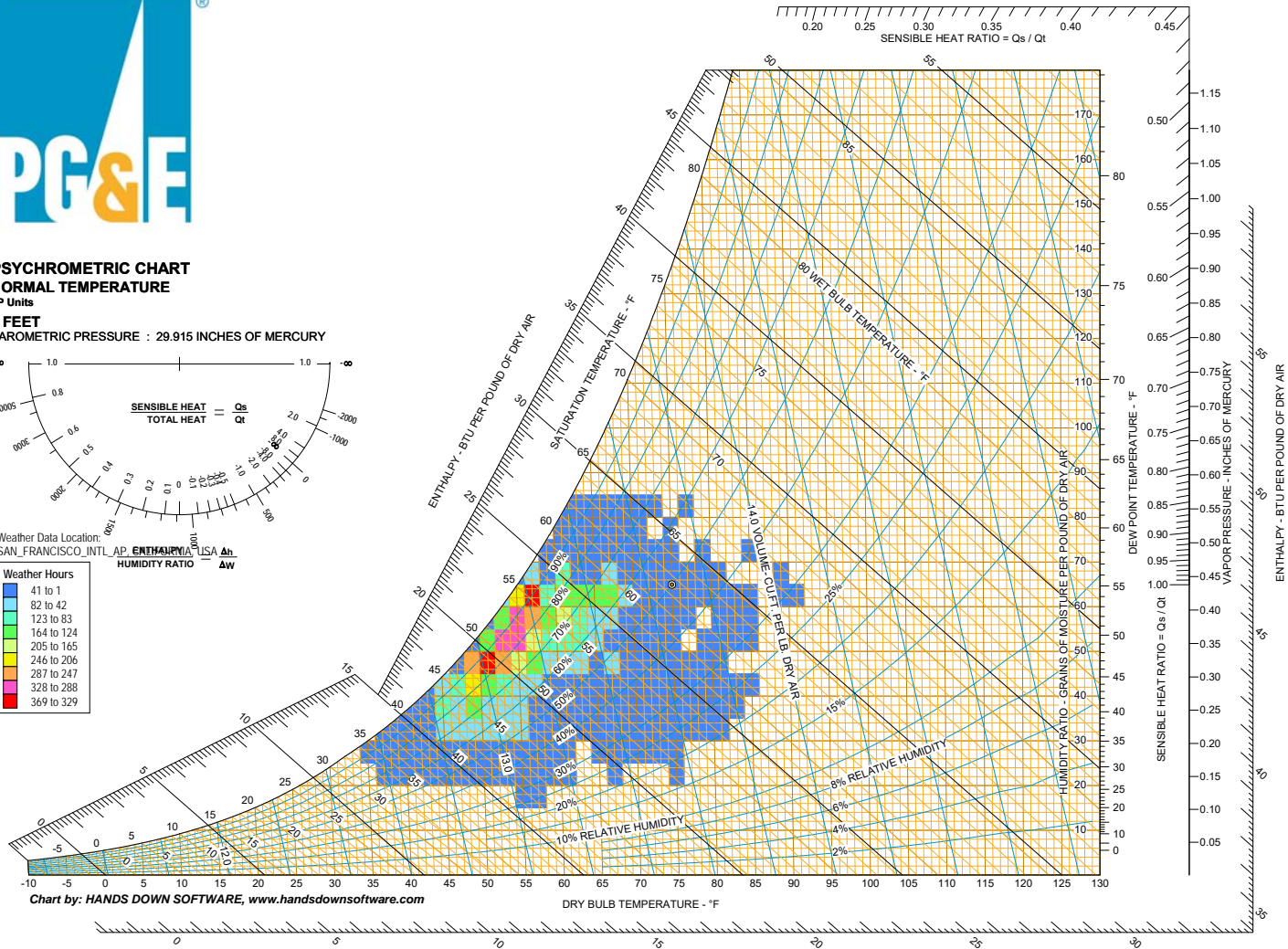
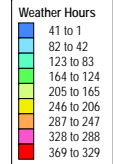
I-P Units

7 FEET

BAROMETRIC PRESSURE : 29.915 INCHES OF MERCURY



Weather Data Location:  
SAN FRANCISCO INTL AP, ENTHALPIA, USA



File Not Saved



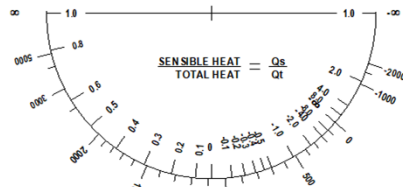


# PSYCHROMETRIC CHART NORMAL TEMPERATURE

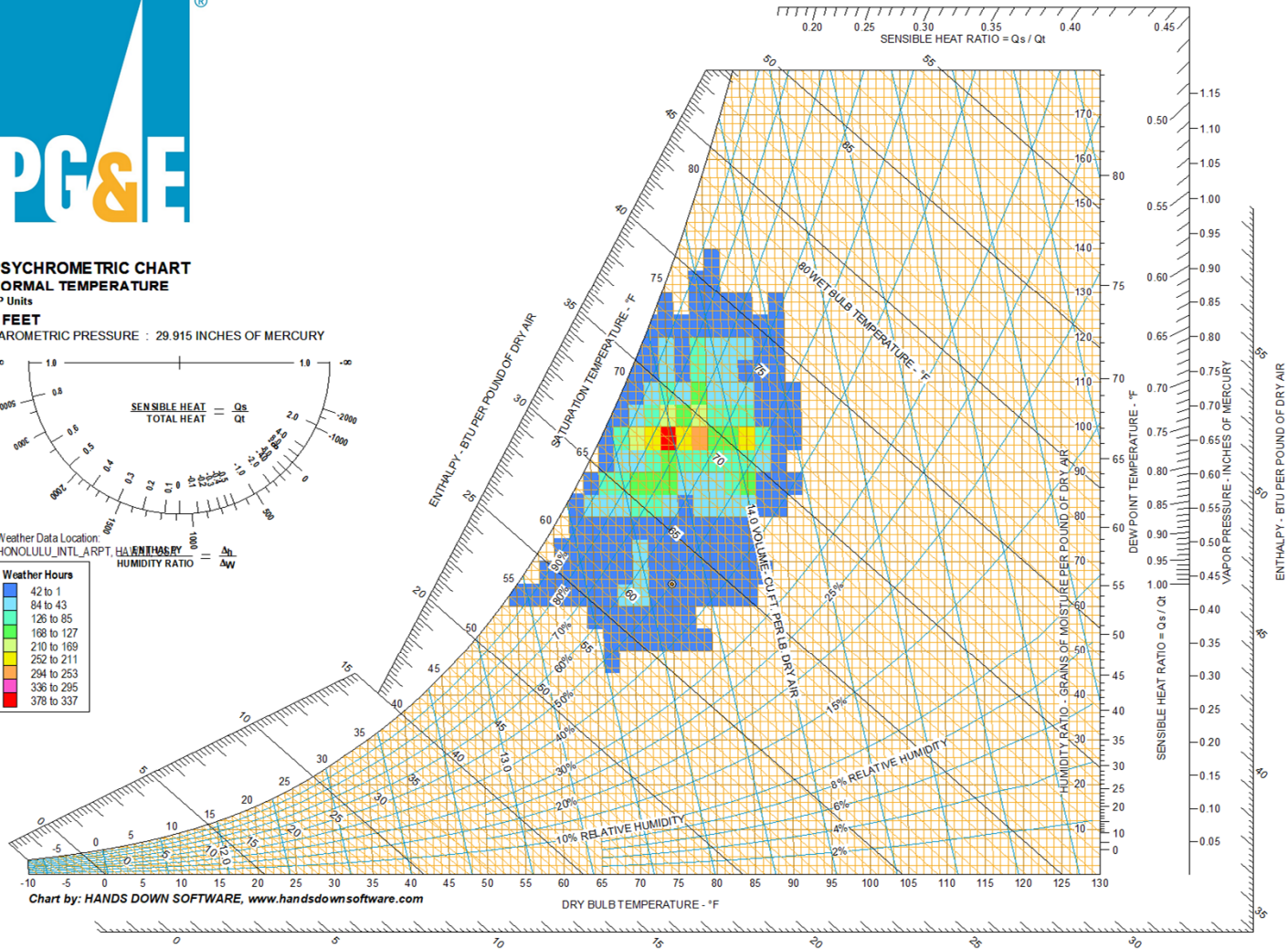
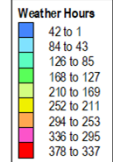
I-P Units

7 FEET

BAROMETRIC PRESSURE : 29.915 INCHES OF MERCURY



Weather Data Location:  
HONOLULU\_INTL\_APT, HAWAII



File Not Saved

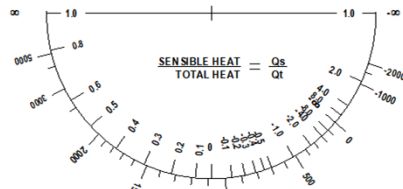


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I-P Units

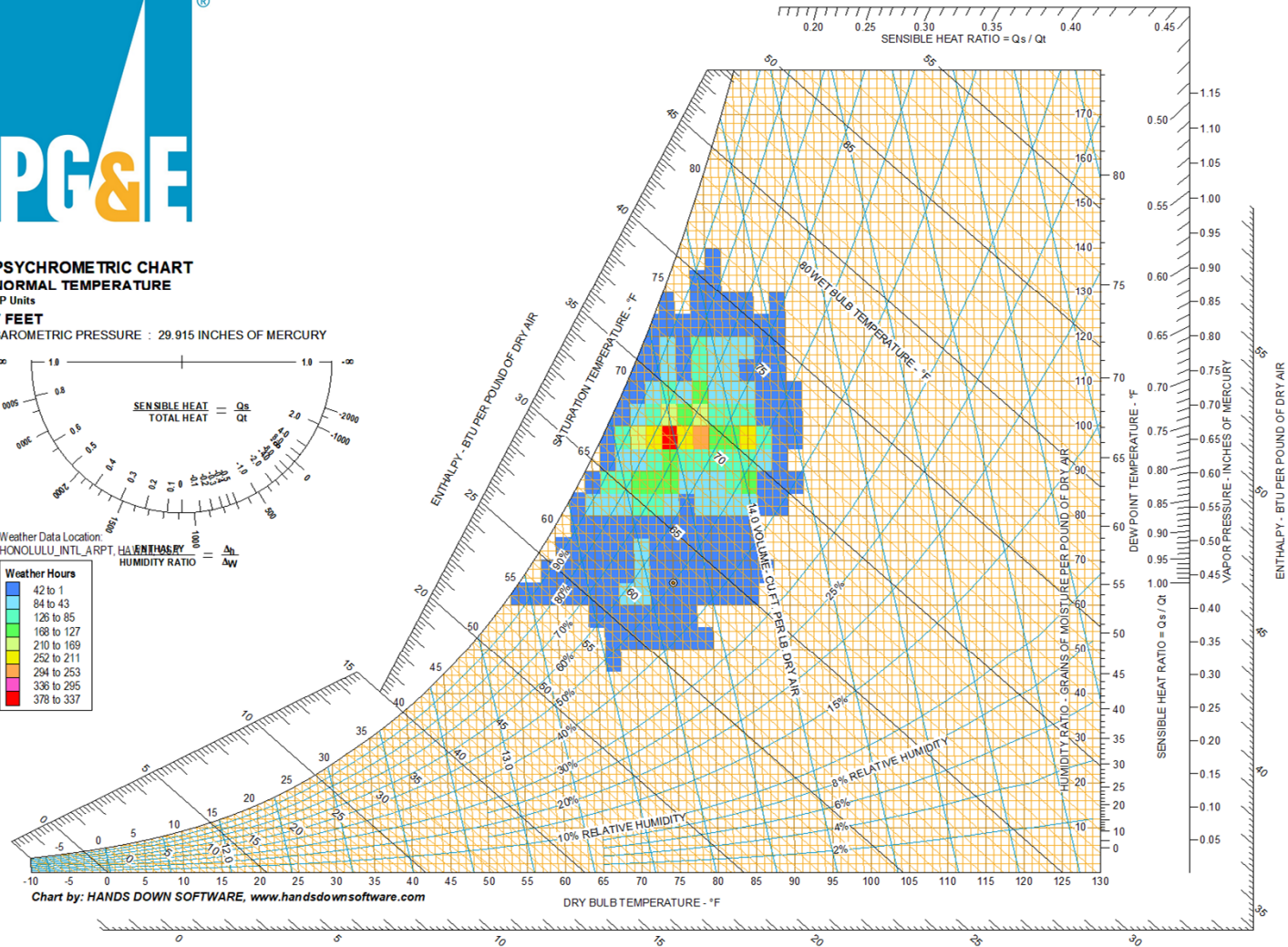
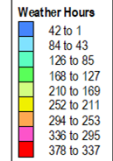
7 FEET

BAROMETRIC PRESSURE : 29.915 INCHES OF MERCURY



Weather Data Location:

HONOLULU\_INTL\_APT, HAWAII



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# You Could Do the Math

The following relationship can be derived from the ideal gas laws by dividing the ideal gas equation for water vapor by the ideal gas equation for air.

$$\frac{p_H}{p - p_H} = w \times \frac{R_H}{R_a}$$

Where:

$p_H$  = The partial pressure of water vapor in the sample in lb/ft<sup>2</sup> absolute.

$p$  = The total pressure of the air and water vapor mixture in the sample in lb/ft<sup>2</sup> absolute.

$w$  = The humidity ratio (humidity ratio is sometimes called specific humidity) in pounds of water per pound of dry air. If it is given in grains per pound, you can convert it to pounds by using the conversion factor of 7,000 grains per pound.

$R_H$  = The gas constant for water vapor (85.8 ft. per°F).

$R_a$  = The gas constant for air (53.3 ft. per°F).

The equations can also be arranged in the following form.

$$w = \frac{R_a}{R_H} \times \frac{p_H}{p - p_H}$$

A special case for saturation.

$$w_s = \frac{R_a}{R_H} \times \frac{p_s}{p - p_s}$$

Where:

$p_s$  = The partial pressure of water vapor in the sample at saturation in lb/ft<sup>2</sup> absolute.

You can also determine the dew point if you know the humidity ratio and the pressure by solving the relationship for the partial pressure of water at saturation and then using a steam table to look up the saturation temperature associated with the partial pressure of water at saturation.

$$w_s = \frac{R_a}{R_H} \times \frac{p_s}{p - p_s}$$

Therefore:

$$w_s \times \frac{R_H}{R_a} = \frac{p_s}{p - p_s}$$

$$\left( w_s \times \frac{R_H}{R_a} \right) \times (p - p_s) = p_s$$

$$p \times \left( w_s \times \frac{R_H}{R_a} \right) - p_s \times \left( w_s \times \frac{R_H}{R_a} \right) = p_s$$

$$p \times \left( w_s \times \frac{R_H}{R_a} \right) = p_s + \left[ p_s \times \left( w_s \times \frac{R_H}{R_a} \right) \right]$$

$$p \times \left( w_s \times \frac{R_H}{R_a} \right) = p_s \times \left[ 1 + \left( w_s \times \frac{R_H}{R_a} \right) \right]$$

# You Could Do the Math

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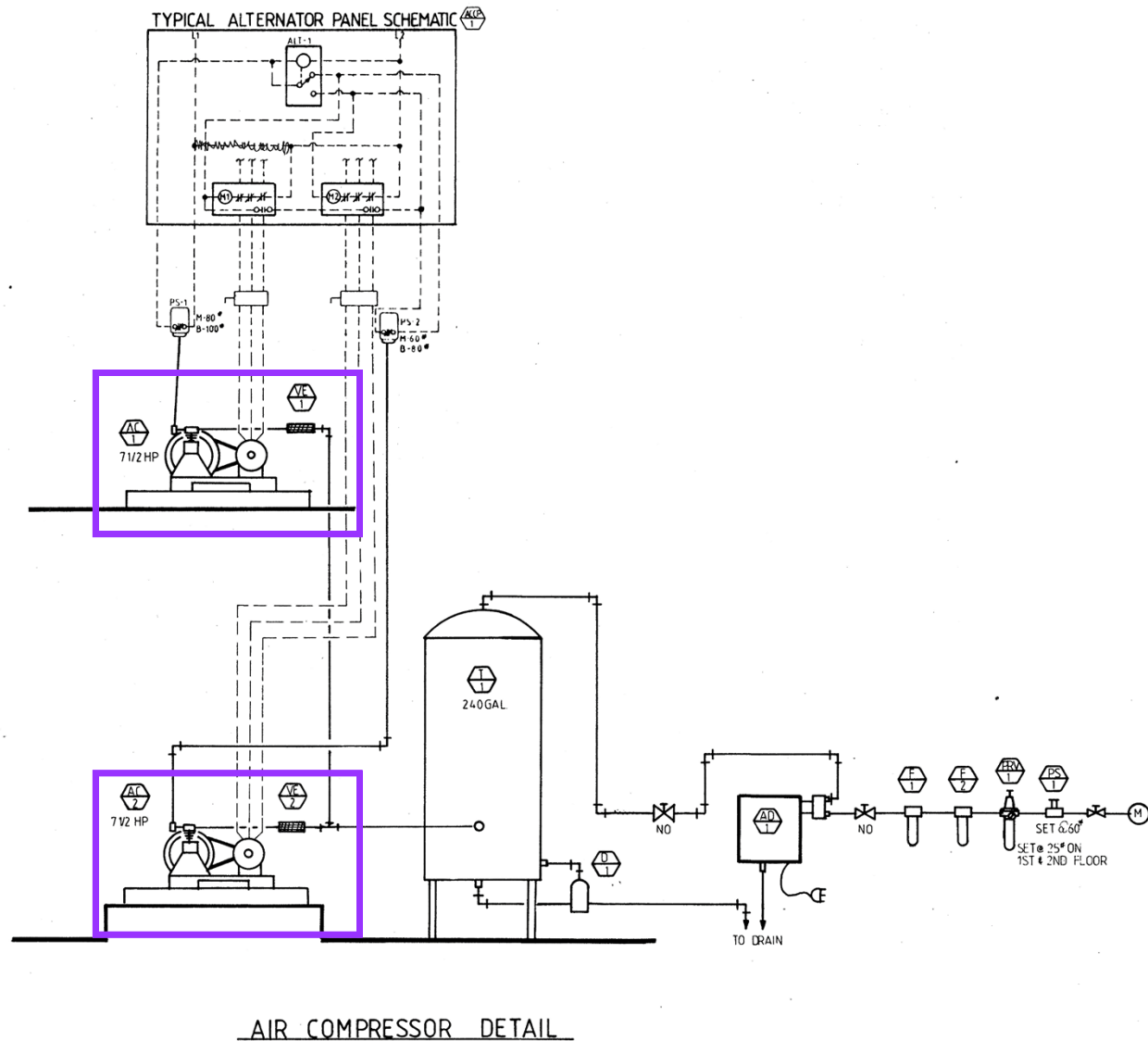
$R_H$  = The gas constant for water vapor (85.8 ft-lb/lbm-°R)



# A Typical Pneumatic Air Supply System

## Key Points

1. Pneumatics are truly “plug and play” Jay Santos
2. Pressure regulation is important
3. Clean, oil-free air is important
4. Dry air is important
5. Redundancy is important



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5. Redundancy is important
6. A tight system is important

